

SORM; F.

7

The synthesis of L-4-amino-3-isoxazolidinone, the unnatural stereoisomer of cycloserine and its antibiotic activity. J. Sautl, J. Beránek, J. Sührig, J. Škoda, V. P. Hess, and E. Kern (Czech. Acad. Sci., Prague). *Experientia* 13, 291 (1957) (in English); *Ch. Abstr.* 51, 1469h. — L-Serine or threonine (Me ester) was successively treated with NEt<sub>3</sub> and thionyl chloride to give 1-(triphenylethyl)-2-carbamoylpyrrolidine (I) (l. m. 123-5°, [α]<sub>D</sub><sup>20</sup> -91.2° in CHCl<sub>3</sub>; n. m. 123-4°, [α]<sub>D</sub><sup>20</sup> 95.0° in CHCl<sub>3</sub>). I was then treated with NH<sub>4</sub>OH and MeONa to yield 1-triphenylethyl-2-hydroxycarbamoylpyrrolidine (II) (l. m. 104°, [α]<sub>D</sub><sup>20</sup> -101.9° in CHCl<sub>3</sub>; n. m. 103°, [α]<sub>D</sub><sup>20</sup> 102.0° in CHCl<sub>3</sub>). II was converted in HCl to α-sulfamoylchloro-N-hydroxypropionamide (III) (l. m. 102-4°, [α]<sub>D</sub><sup>20</sup> -14.1° in MeOH; n. m. 101-5°, [α]<sub>D</sub><sup>20</sup> 14.2° in MeOH). III was then cyclized with Zn dust HF to L-4-amino-3-isoxazolidinone (l. m. 153-4°, [α]<sub>D</sub><sup>20</sup> -114.5° in H<sub>2</sub>O; n. m. 152-3°, [α]<sub>D</sub><sup>20</sup> 115° in H<sub>2</sub>O). Tests with cultures of *Escherichia coli* indicate that D-4-amino-3-isoxazolidinone is substantially more antibiotic than either of the optically active forms. D. S. Farner

GRYUNBERGER, D. [Grünberger, D.]; SHORMOVA, Z. [Šormova, Z.]; SHORM, F.  
[Šorm, F.]

Effect of albomycin on oxidation processes and nucleic acid metabolism  
in Staphylococcus aureus and Escherichia coli [with summary in  
German]. Biokhimiia 22 no.1/2:148-153 Ja-F '57. (MIRA 10:7)

1. Biokhimicheskoye otdeleniye Khimicheskogo instituta Československoy  
akademii nauk, Praga.

(MICROCOCCUS PYOGENES, effect of drugs on,  
albomycin, on oxidation & nucleic acid metab. (Rus))

(ESCHERICHIA COLI, effect of drugs on  
same)

(ANTIBIOTICS, effects,  
albomycin on E.coli & Micrococcus pyogenes aureus  
oxidation & nucleic acid metab. (Rus))

(NUCLEIC ACIDS, metabolism,  
E.coli & Micrococcus pyogenes aureus, eff. of albomycin  
(Rus))

SCRM, F.; ROMANUK, M.

"Terpenes. LXXIII cis- and trans-bicyclo (o, 3, 5) decane. In German."

p. 779 (Collection of Czechoslovak Chemical Communications, Sbornik, Chekhoslovatskikh Khimicheskikh Rabot) Vol. 22, no. 3, June 1957  
Prague, Czechoslovakia

SO: Monthly Index of East European Accessions (EEAI) LC. Vol. 7, no. 4,  
April 1958

SECRET  
CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring Substances  
and Their Synthetic Analogs. G-3

Abs Jour: Referat Zhur-Khimiya, No 4, 1958, 11420.

Author : Jarolim, V., Streibl, M., Dolejs, L., and Sorm, F.

Inst : Not given.

Title : On Terpenes. LXXV. Cis- and Trans-homocaryophyllenic  
Acid. LXXVI. Synthesis of 4,8,11,11-tetramethylricyclo-  
(0,2,7)-undecane (Caryophyllene).

Orig Pub: Sbornik Chekchoslov Khim Rabot, 22, No 4, 1266-1276;  
1277-1282 (1957) (in English with a summary in Russian)

Abstract: See RzhKhim, 1957, 44661, 44662.

Card : 1/1

3

CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring Substances  
and their Synthetic Analogs. G-3

Abs Jour: Referat Zhur-Khimiya, No 4, 1958, 11457.

Author : Smrt, J., Beranek, J., Sicher, J., and Sorm, F.

Inst :

Title : Synthesis of 4-amin-3-isoxazolidene (Cycloserine)

Orig Pub: Chem Listy, 51, No 1, 112-122 (1957) (in Czech);  
Sbornik Chekchoslov Khim Fabot, 22, No 1, 262-273  
(in English with a summary in Russian)

Abstract: The antibiotic cycloserine (I) (see RZhKhim, 1956, 16239) has been synthesized from the methyl ester of N-tritylserine (II) via the methyl ester of O-mesyl-N-tritylserine (III), 1-trityl-2-carbomethoxyethyleneimine (IV), which on reaction with  $\text{NH}_2\text{OH}\cdot\text{HCl}$  gives

Card : 1/10

CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring Substances  
and their Synthetic Analogs. G-3

Abs Jour: Referat Zhur-Khimiya, No 4, 1958, 11457.

hydroamic acid (V); the latter adds HCl with the formation of the hydrochloride of  $\alpha$ -amino- $\beta$ -chloropropionhydroamic acid (VI); strongly basic anion exchange resins cyclize VI to I. For comparison purposes 4-benzylamine- (IX) and 4-benzhydrylaminoisoxazolidone-3 (X) were synthesized from N-benzyl-2-carbomethoxyethyleneimine (VII) and N-benzhydryl-2-carbomethoxyethyleneimine (VIII) by the same method. 1-benzylethyleneimine-2-carbhydroamic acid (XI) is synthesized by refluxing 87.5 gms of the methyl ester of 1,2-dibromopropionic acid [sic] in 550 ml  $\text{C}_6\text{H}_6$  for 3 hrs with 71.4 gms triethylamine and 38.2 gms benzylamine, shaking the mixture with water for 12 hrs and allowing the mixture to stand with VII [TN: meaning appears garbled] for 48 hrs, obtained by eva-

Card : 2/10

CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring Substances  
and their Synthetic Analogs.

G-3

Abs Jour: Referat Zhur-Khimiya, No 4, 1958, 11467.

0.5 liter  $\text{CH}_3\text{OH}$  with 12.5 gms  $\text{Na}_2\text{CO}_3$  in 1 liter water, allowing to stand 4 days, and acidifying with 50 ml  $\text{CH}_3\text{COOH}$  in 200 ml alcohol; the yield is 73%, mp 139-141° (decomp: from  $\text{CH}_3\text{OH}$ ). Hydrogenation and hydrolysis of the latter product give serine. The hydrogenation of 0.03 mol X in 100 ml alcohol and 1 ml  $\text{CH}_3\text{COOH}$  over  $\text{PtO}_2$  by 200 ml  $\text{H}_2$  gives 2.15 gms of the amide of N-benzhydrylserine, mp 142-144° (from alcohol). A mixture of 0.03 mol II, 50 ml dry pyridine, and 2.5 ml mesyl chloride is kept 24 hrs at 0.4°, diluted with 200 ml water and  $\text{CHCl}_3$ ; III is extracted in 90% yield, mp 123° (from benzene-alcohol) 0.01 mol II in 50 ml dioxane is mixed with  $\text{NH}_2\text{OH}$  (prepared from 14 gms  $\text{NH}_2\text{OH}\cdot\text{HCl}$  in 100 ml abs

Card : 7/10

CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring Substances and their Synthetic Analogs.

G-3

Abs Jour: Referat Zhur-Khimiya, No 4, 1958, 11467.

$\text{CH}_3\text{OH}$  and 6.9 gms Na in 80 ml  $\text{CH}_3\text{OH}$ ; after 3 days the reaction mixture is diluted with 100 ml water, and neutralized with  $\text{CH}_3\text{COOH}$ ; further dilution results in the formation of a precipitate (33 gms) of N-tritylserinehydroxamic acid, mp 109° ( $\text{CH}_3\text{OH}$ ); the product contains 1 molecule of combined  $\text{CH}_3\text{OH}$ . A mixture of 0.02 mol III, 20 ml  $\text{C}_6\text{H}_6$ , 1 ml  $\text{CH}_3\text{CH}_3$ , and 2.5 gms N-ethylpiperidine is refluxed 8 hrs, diluted with 15 ml  $\text{CHCl}_3$ , washed with water, and evaporated; the yield of IV is 80%, mp 130-131° (from benzene-cyclohexane). On standing for 3 days a mixture of 0.18 mol IV in 100 ml dioxane and 25.2 gms  $\text{NH}_2\text{OH}\cdot\text{HCl}$  and 12.4 gms Na in 150 ml  $\text{CH}_3\text{OH}$

Card : 8/10

Abs Jour: Referat Zhur-Khimiya, No 4, 1958, 11467.

gas; after 3 hrs VI is obtained by suction filtration, yield 66%, mp 191° (decomp; from CH<sub>3</sub>OH-ether). Hydrogenation and hydrolysis of the latter product give XII. 0.06 mol VI in 100 ml water is passed for 10 min through a column containing 300 ml [sic] of amberlite IRA-4000 anion exchange resin (strongly basic), the column is washed with 1 liter water (1 hr), and the product is eluted at 0° with 20% CH<sub>3</sub>COOH. The eluate is collected until the pH attains 5.8, and the solution (60 ml) is diluted with 450 ml alcohol; at -50° DL-I precipitates, yield 51%, mp 141-142.5° (from 20% alcohol).

Card      ; 10/10

CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring Substances and Their Synthetic Analogs.

G-3

Abs Jour: Ref Zhur-Khim., No 13, 1958, 43491.

Author : Fajkos J., Sorn F.

Inst :

Title : On Steroids. XXVIII. Synthesis of Analogues of Progesterone and Desoxycorticosterone With the Side Chain in Position 16.

Orig Pub: Sb. chekhosl. khim rabot, 1957, 22, No 6, 1873-1886.

Abstract: See RZhKhim, 1958, 11432.

Card : 1/1

CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring Substances and Their Synthetic Analogs.

G-3

Abs Jour: Ref Zhur-Khim., No 13, 1958, 43474.

Author : Suchy M., Herout V., Sorn F.

Inst :

Title : On Terpenes. LXXVII. On the Nature of Arctiopicrin -- the Unsaturated Lactone from Arctum Mines Bernh.

Orig Pub: Sb. chekhosl. khim. rabot, 1957, 22, No 6, 1902-1908.

Abstract: See RZhKhim, 1957, 54490.

Card : 1/1

Soll, F.

"Academician R. Lukes at sixty." p. 399.

Institute of Applied Physics, (Czechoslovak Academy of Sciences.) Vol. 51, no. 3,  
Mar. 1957.

EAST

SU: Monthly Index of European Accession (EEAI) LC, Vol. 7, No. 5 May 1958

SORM, FRANTISEK

CZECHOSLOVAKIA/Organic Chemistry - Synthetic Organic Chemistry. G-2

Abs Jour: Referat Zhur-Khimiya, No 5, 1958, 14358.

Author : Novak Jiri, Ratusky Josef, Smeberk Vladimir, Sorm Frantisek.

Inst :  
Title : Reactions of Diazoketones. I. Interaction of Diazoacetone with Unsaturated Compounds.

Orig Pub: Chem. listy, 1957, 51, No 3, 479-492; Sb. chekhosl. khim. rabot, 1957, 22, No 6, 1836-1851.

Abstract: Diazoketones [diazoacetone (I)] react with unsaturated compounds to form derivatives of acetyl cyclopropane. Reaction with aromatic compounds RH results in ketones  $RCH_2COCH_3$  (II);  $C_6H_6$  reacts only in the presence of  $BF_3$ ; the yield of phenylacetone is 4%. Vinyl esters and ethers usually yield a mixture of cyclopropane derivatives and ketones II. Acrylonitrile forms with I (1 hour,  $100^\circ$ . addition of Cu) a compound

Card : 1/5

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CZECHOSLOVAKIA/Organic Chemistry - Synthetic Organic Chemistry. G-2

Abs Jour: Referat Zhur-Khimiya, No 5, 1958, 14358.

$C_9H_{11}O_2N_2$ , BP  $120^\circ/0.3$  mm, MP  $95^\circ$ , the structure of which was not investigated. Furan and sylvan are cleaved by I. To 15 g styrene and 1 g Cu are added at  $100^\circ$  5 g styrene + 3 g I and mixture is heated for 2 hours; 1-acetyl-2-phenylcyclopropane (III) is distilled and subjected to chromatography on  $Al_2O_3$ , yield 42%, BP  $73^\circ/0.3$  mm; semicarbazone (SC), MP  $135^\circ$ . III is hydrogenated, over Pd/C, to 1-phenylpentanone-4, BP  $130^\circ/17$  mm,  $64^\circ/0.22$  mm; SC, MP  $128^\circ$ ; oxidation of III with a solution of  $Br_2$  in aqueous NaOH yields 2-phenylcyclopropanecarboxyl-1 acid, MP  $90^\circ$ ; amide, MP  $189^\circ$ . Similarly to III there is formed from I and *ortho*-ethylstyrene, in cyclohexane at  $85^\circ$  in the presence of  $CoSO_4$ , 1-cetyl-2-phenyl-3-ethylcyclopropane, BP  $78^\circ/0.005$  mm. Solution of 5 g I in cyclohexane is added to boiling suspension of 0.2 g anhydrous  $CuSO_4$  in cyclohexane, to get 38.7% of 7-acetyl-bicyclo-[4,1,0]-heptane, BP  $83-86^\circ/10$  mm,  $n_D^{20}$  1.4805; SC, MP  $175^\circ$ ;

Card : 2/5

CZECHOSLOVAKIA/Organic Chemistry - Synthetic Organic Chemistry.

G-2

Abs Jour: Referat Zhur-Khimiya, No 5, 1958, 14358.

only the acetate of 7-acetyl-bicyclo-[4,1,0]-heptanol-1, yield 12%, BP 72-76°/0.3 mm; SC, MP 184°. In the same manner from dihydropyran was obtained only 7-acetyl-2-oxabicyclo-[4,1,0]-heptane (IV), yield 54%, BP 80-83°/10 mm; MP 144. IV was obtained also from the acid chloride of 2-oxabicyclo-[4,1,0]-heptane-7-carboxylic acid, MP 89-93°/15 mm (V--acid) and (CH<sub>3</sub>)<sub>2</sub>Cd in C<sub>6</sub>H<sub>6</sub>, yield 68%. V was obtained from dihydropyran and N<sub>2</sub>CHCOOC<sub>2</sub>H<sub>5</sub> with subsequent saponification with NaOH, yield 86%, MP 104°. Indole reacts with I in cyclohexane with addition of Cu at 80°; by distillation of the mixture and chromatographic purification is isolated, with a yield of 25%, 3-indolylacetone (MP 116.5-118.5°; picrate, MP 116-117°), which is reduced with LiAlH<sub>4</sub> to 1-(3-indolyl)-propanol-2, yield 89%, BP 146-147°/0.35 mm, MP 37.5°. Coumarone and CuSO<sub>4</sub> give with I at 80° 6-acetyl-2-oxa-3,4-benzobicyclo-[3,1,0]-hexene-3, MP 63°, which is converted with NaOBr in aqueous dioxane to 2-oxa-3,4-benzobicyclo-

Card : 4/5

CZECHOSLOVAKIA / Organic Chemistry. Natural Substances and  
Their Synthetic Analogues.

G-3

Abs Jour : RZhKhim., No 10, 1958, 32586

Author : Vlastimil Herout, Ladislav Dolejs, Frantisek Sorm

Inst : Not given

Title : Terpenes. LXXIX. Structure of Artabsine, Prochamazulen-  
ogenetic Substance from Bitter Absinth.

Orig Pub : Chem. listy, 1957, 51, No 3, 572-578; Sb. chekhosl. khim.  
rabot, 1957, 22, No 6, 1914-1920

Abstract : The authors attribute the probable structure of 4-oxy-  
guaianodiene-2,4(10)-olide-8,12 to artabsine (I), the  
prochamazulenogenetic substance from bitter absinth  
(*Artemisia absinthium* L.) (see RZhKhim, 1955, 34500),  
based on the hydrogenation and oxidation products. Their  
reactions and infra-red spectra. But the structure of 4-o-  
xy-guaianodiene-1,3-olide-8,12 (RZhKhim, 1957, 37762) cannot

Card 1/4

34

CZECHOSLOVAKIA / Organic Chemistry. Natural Substances and  
Their Synthetic Analogues.

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CIA-RDP86-00513R00165242001

G-3

Abs Jour : RZhKhim., No 10, 1958, No 32586

be completely excluded. 1.63 g of 1-oxyguaiaene-4(10)-  
-olide-8,12 (dihydroartabsine) (II), melting point 133.5  
to 134° (from isopropyl ether),  $[\alpha]_D^{20} = -13.6^{\circ}$  (c = 1.71,  
all  $\alpha$ -s in chloroform), was prepared by hydrogenation of  
2.13 g of I in 10 mlit of alcohol on 0.17 g of PtO<sub>2</sub> and  
chromatography on 170 g of Al<sub>2</sub>O<sub>3</sub> in C<sub>6</sub>H<sub>6</sub>. 110 mg of oxy-  
oxylactone (III), melting point 160° (from alcohol - iso-  
propyl ether), was produced by ozonization of 150 mg of II  
in 5 mlit of CH<sub>3</sub>COCH (30 min. at 15°) and hydrogenation of  
the product on 50 mg of PtO<sub>2</sub>. Ketooxylactone (IV) C<sub>15</sub>H<sub>22</sub>O<sub>4</sub>,  
melting point 145° (from isopropyl ether), yield 20 mg, was  
produced from the neutral fraction by oxidation of 33 mg of  
III by 50 mg of CrO<sub>3</sub> in 2 mlit of glacial CH<sub>3</sub>COCH (12 hours  
at 20°). 4 products were obtained by hydrogenation of 5.66  
g of I in 10 mlit of glacial CH<sub>3</sub>COCH on 692 mg of PtO<sub>2</sub> and

Card 2/4

Abs Jour : RZhKhimi, No 10, 1958, No 32586

CrO<sub>3</sub> does not oxidize VI "a", "b" and "c" into CH<sub>3</sub>COOH. 70 mg of guaiacene-*i*-olide-8,12 (VII), boiling point 128 to 129° under 1.2 mm was received by dehydrogenation of 100 mg of VI "a" with 150 mg of SOCl<sub>2</sub> in 1.5 mlit of pyridine (15 hours at 0° and 2 hours at 20°). 320 mg of trioxy-1,4-oxidoguaienolide-8-12, melting point 188° (from alcohol - isopropyl ether), was obtained from the neutral fraction by oxidation of 1 g of I with 1.5 g of KMnO<sub>4</sub> in the solution of 100 mlit of acetone, 4 mlit of water and 1 mlit of CH<sub>3</sub>COOH (1.5 hour at 20°). HCOOH is formed at the oxidation of the product with NaIO<sub>4</sub> in water. The infrared spectra of IV, V, VI "a", "b" and "c" and VII are presented.

Card 4/4

SORM, F.

CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring Substances and Their Synthetic Analogs.

G-3

APPROVED FOR RELEASE: 08/25/2000, CIA-RDP86-00513R001652420016-4"

Author : Fajkos, J. and Sorm, F.

Inst :

Title : On Steroids. XXVIII. Synthesis of Analogs of Progesterone and Desoxycorticosterone with the Side Chain in the 16-Position.

Orig Pub: Chem Listy, 51, No 3, 579-591 (1957) (in Czech)

Abstract: The following analogs of progesterone and desoxycorticosterone having the side chain in the 16-position have been synthesized: the action of CH<sub>3</sub>MgBr on 3 $\beta$ -acetoxy-16-cyanoandrostadiene-5,16 (II) which is selectively hydrogenated to 3 $\beta$ -hydroxy-16 $\beta$ -

Card : 1/10

6

CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring Substances and Their Synthetic Analogs.

G-3

Abs Jour: Referat Zhur-Khimiya, No 4, 1958, 11432.

CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring  
Substances and their Synthetic Analogs.

G-3

Abs Jour: Referat Zhur-Khimiya, No 4, 1958, 11432.

( $c = 1.77$ ; alcohol). When a suspension of 0.3 gm XIII in 1.2 ml oxalyl chloride and 12 ml  $C_6H_6$  is shaken for 15 min and evaporated under vacuum, 360 mg XV are obtained; the latter are dissolved in 6 ml  $C_6H_6$  and poured at  $-10$  to  $0^\circ$  into a solution of  $CH_2N_2$  (from 5 gm nitrosomethylurea); after 1 hr the filtrate is evaporated and the XVI which is obtained (340 mg) is heated (1 hr,  $90^\circ$ ) with 1.5 ml 5%  $CH_3COONa$  in  $CH_3COOH$ , the solution is evaporated, and the oily product is extracted with ether; chromatography on  $Al_2O_3$  gives 14% XVII (elution with petroleum ether : benzene = 1 : 2), mp  $194-195^\circ$  (from acetone),  $[\alpha]_D^{20} + 36 \pm 1^\circ$  ( $c = 2.65$ ). When 1 gm XX is refluxed for 1.5 hrs with 0.3 gm  $K_2CO_3$  in 50 ml  $CH_3OH$  and 10 ml water, an 83% yield of 3-hydroxy-17-

Card : 8/10

CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring  
Substances and their Synthetic Analogs.

G-3

"APPROVED FOR RELEASE: 08/25/2000 CIA-RDP86-00513R001652420016-4"

Abs Jour: Referat Zhur-Khimiya, No 4, 1958, 11432.

cycloandrosta-5,16 is obtained, mp  $170-171^\circ$  (from  $CH_3OH$ ),  $[\alpha]_D^{20} - 66 \pm 1^\circ$  ( $c = 3.18$ ); the latter product is oxidized (as in the case of II to VIII) to 3-keto-17-cyanoandrosta-4,16 (XXVIII), yield 68%, mp  $156-158^\circ$  (from alcohol)  $[\alpha]_D^{20} + 168 \pm 2^\circ$  ( $c = 2.08$ ). When a mixture of 325 mg XXVIII, 2 ml ethylene glycol, 15 mg p-toluenesulfonic acid, and 10 ml  $C_6H_6$  is refluxed for 20 hrs (removal of water), a 76% yield of 3-ethylenedihydroxy-17-cyanoandrosta-5,16 is obtained, mp  $217-219^\circ$  (from alcohol),  $[\alpha]_D^{20} - 45 \pm 1^\circ$  ( $c = 2.45$ ) which is converted to XXI (as in the case of I to II, decomposition with a solution of  $NH_4Cl$ ); after purification by chromatography on  $Al_2O_3$  the yield

Card : 9/10

10

CZECHOSLOVAKIA/Organic Chemistry - Theoretical and General  
Questions on Organic Chemistry.

G.

Abs Jour : Ref Zhur - Khimiya, No 9, 1958, 28643

later becomes very marked. In the opinion of the authors methyl radicals are present in the reaction space both in the gas phase and on the surface of the Si or at the surface of the contact mass (CM). The radicals in the gas phase combine to form silicon-free reaction products while the radicals localized on the surface of the Si or on the CM form the methylchlorosilanes proper. The gradual decrease in the yield of methylchlorosilanes continues until the radicals present in the gas phase are exhausted. The sharp decrease in yield begins when the NO which is added begins to react also with the radicals localized on the surface of the CM. The reactions of the methyl radicals in the gas phase are discussed, in particular, the reactions with  $\text{CHCl}_3$  and  $\text{H}_2$ , thermal decomposition,

Card 2/3

APPROVED FOR RELEASE: 08/25/2000 CIA-RDP86-00513R001652420016-4"

CZECHOSLOVAKIA/Organic Chemistry - Theoretical and General  
Questions on Organic Chemistry.

J.

Abs Jour : Ref Zhur - Khimiya, No 9, 1958, 28643

and disproportionation reactions. The theoretical conclusions are compared with the experimental data on the composition of the reaction products (see RZhKhin, 1957, 68912).

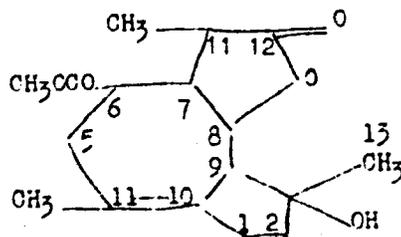
Card 3/3

CZECHOSLOVAKIA / Organic Chemistry. Natural Substances and  
Their Synthetic analogues.

G-3

Abs Jour : RZhKhim., No 10, 1958, No 32587

ducts, 6-acetoxycucurbitolides-8,12 (IIa and IIb) with melting points  $115.5^{\circ}$  and  $123^{\circ}$ , were produced by hydrogenation of I on Pt (from  $PtO_2$ ) in glacial  $CH_3COCH_3$ , this indicates that the oxy group in I is bonded with the carbon atom next to a binary bond. 1-oxy-6-acetoxycucurbitolide-8,12 (tetrahydro-matricine) (III) together with a little amount of dihydro-matricine (IV) was produced by hydrogenation of I on  $PtO_2$  in alcohol. IIa when saponified with  $K_2CO_3$  in  $CH_3OH$  produces the corresponding 6-oxycucurbitolide-8,12(V).



Card 2/8

CZECHOSLOVAKIA / Organic Chemistry. Natural Substances and  
Their Synthetic Analogues.

G-3

Abs Jour : RZhKhim., No 10, 1958, No 32587

totrol (XII) was detected only in a small amount in mother liquors (by paper chromatography). XII was synthesized by reducing with  $\text{LiAlH}_4$  at an elevated temperature in N-ethylpyridine. XI produces X when dehydrogenated with Se, while XII produces a mixture of X with linderazulene (XIII) under the same conditions. The formation of XIII together with X from XII is a proof that the position of the acetoxy group is at C(6). III is stable with respect to  $\text{CrO}_2$  in glacial  $\text{CH}_3\text{COOH}$  in the cold, while VI produces 1-oxy-6-ketoguaianolide-8,12 (XIV) under the same conditions, which is a proof that the free oxy group in I is a tertiary group. Koto-guaianone-1(9)-olide-8,12 (XV) was synthesized by the dehydrogenation of XIV with  $\text{BF}_3$  esterate or  $\text{HCOOH}$ , its binary bond is not conjugate either with the lactone carbonyl, or the ketone carbonyl, which is shown by the infrared and

Card 4/8

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Organic Chemistry. Natural Substances and  
Their Synthetic Analogues.

G-3

CIA-RDP86-00513R001652420016-4

Abs Jour : RZhKhim., No 10, 1958, No 32587

ultraviolet spectra. XII does not change even after a 70 hour action of  $\text{HIO}_4$ . The above presented formula was attributed based on a discussion of ultraviolet and infrared spectra and all the experimental results. IIa and IIb, melting point  $115.5^\circ$ ,  $[\alpha]_D^{20} = -6.5^\circ$  (c = 10.1; all  $\alpha$ -s in chloroform); melting point  $123^\circ$ ,  $[\alpha]_D^{20} = -9.0^\circ$  (c = 4.3). III, melting point  $183$  to  $184^\circ$ ,  $[\alpha]_D^{20} = +12.5^\circ$  (c = 6.8); IV, melting point  $166.6$  to  $168^\circ$ ,  $[\alpha]_D^{20} = -20.5^\circ$  (c = 11.1), was separated from mother liquors after the crystallization of III. IV can be converted into III by hydrogenation on  $\text{PtO}_2$  in glacial  $\text{CH}_3\text{COOH}$ . V, melting point  $157^\circ$  (from diisopropyl ether - petroleum ether), was obtained from 289 mg of IIa and aqueous-methanol solution of  $\text{K}_2\text{CO}_3$  (48 hours of seasoning), yield 221 mg. Similarly, 0.85 g of VI, melting point  $138.5$  to  $140^\circ$  (from diisopropyl ether - acetone),  $[\alpha]_D^{20} = +33.0^\circ$  (c = 0.91), was received from 1.1 g of III.

Card 5/8

Abs Jour : RZhKhim., No 10, 1958, No 32587

= 0.92), was prepared of III by reducing it with  $\text{LiAlH}_4$  in ether. The reduction of 300 mg of III with 1.2 g of  $\text{LiAlH}_4$  in boiling N-ethylpiperidine (4 hours) resulted in XII, yield 245 mg, melting point 138 to 139° (from diisopropyl ether - acetone),  $[\alpha]_D^{20} = -11.9^\circ$  (c = 0.84). A mixture of X and XIII (1 : 1), melting point 103 to 105° (from alcohol), is obtained by the dehydrogenation of XII with Se; trinitrobenzoate, melting point 150 to 152°. The latter was proved by paper chromatography,  $R_f$  for X is 0.75 (at the application of 45% aq.  $\text{H}_2\text{PO}_4$ ) and  $R_f$  for XIII is 0.30. XIV, melting point 187.5 to 189° (from diisopropyl ether - acetone), was synthesized of 300 mg of VI and 83 mg of  $\text{CrO}_3$  in glacial  $\text{CH}_3\text{COOH}$  (40 hours, 0°), yield 268 mg. XV, melting point 100 to 101° (from diisopropyl ether), was prepared of 300 mg of XIV and 5 mlit of  $\text{BF}_3$  esterate (15 min. seasoning), yield

Card 7/8

39

CZECHOSLOVAKIA / Organic Chemistry, Natural Substances and Their Synthetic Analogues.

G-3

Abs Jour : RZhKhim., No 10, 1958, No 32587

APPROVED FOR RELEASE: 08/25/2000 CIA-RDP86-00513R001652420016-4

226 mg. A method of descending paper chromatography in the system diisopropyl ether - water was developed for the study of the degradation reaction course. The development was carried out with a saturated solution of  $\text{SbCl}_3$  in  $\text{CHCl}_3$ , to which 20% of  $\text{SCCl}_2$  was added. The spots were seen best after the chromatogram was heated to from 90 to 100°. The values of  $R_f$  are: I - 0.64, III - 0.63, VI - 0.25, XI - 0.29, XII - 0.23, XIV - 0.52, and XV - 0.74.

Card 8/8

ARNOLD, Zdenek <sup>1st</sup> Serm, Frantisek

from aq. soln. *H*, m. 188-90° (from pyridine), 35% (picrate, m. 141.5-2.5°); *Me*, m. 178-90° (from  $\text{CH}_2\text{Cl}_2$ ), 20% (picrate, m. 127-8°); *Et*, m. 169.5-71.5° (from pyridine), 30% (picrate, m. 98-9°); *C<sub>6</sub>H<sub>5</sub>* (picrate, m. 113.5-14.5°); *Ph*, m. 236° (from dioxane-MeOH), 25.5° (picrate, m. 114.5-15.5°) (all picrates from EtOH). Depending on the method of isolation, derivs. were prepd. as follows. Passing a slow stream of  $\text{COCl}_2$  into 3.65 g.  $\text{HCONMe}_2$ , until all solidified, adding dropwise 4.35 ml.  $\text{FeCl}_3(\text{OEt})_2$ , heating the mixt. slowly to 75°, keeping 15 min. at 75°, decomp. with ice, and adding 8.2 g.  $\text{AcONa}$  and 12.5 ml. 4*N*  $\text{PhNH}_2\text{HCl}$  gave 4.7 g. cryst. ppt. of  $\text{PhNHCH:CEtCHO:NPh.HCl}$  m. 200-7° (from EtOH). A similar procedure with  $\text{MeCH(OEt)}_2$  combined with pptn. of the product with 86%  $\text{HClO}_4$  gave 74% brown ppt. of  $\text{MeNCH:CH:CH:NPh.HClO}_4$ , m. 158.5° (from EtOH). When 3.65 g.  $\text{FeCH(OEt)}_2$  was treated with excess  $\text{COCl}_2$  and the products worked up as described, the mother liquors after sepn. of II (*R'* = Et) yielded, on extrn. with ligroine and distn., 2 g. yellowish liquid, b<sub>3</sub> 102-5°, *n<sub>D</sub><sup>20</sup>* 1.4753, apparently  $\text{EtOCH:CEtCHO}$  (III), besides 1.3 g.  $\text{Me}_2\text{C:CEtCHO}$ . III shaken with 30% aq.  $\text{NHMe}_2$  gave I (*R* = Et). III was also obtained in 1.3-g. yield by boiling 1.3 g. Na salt of  $\text{EtCH(CHO)}$ , 9 hrs. in 20 ml. EtOH with excess of EtBr. The I heated to 70° with 50% NaOH and the resulting solns. evapd. almost to dryness *in vacuo* gave in 90% yield Na salts of dialdehydes which were pptd. with EtOH-Me<sub>2</sub>CO, dried and converted to the following free  $\text{R''CH(CHO)}$  (*R''* shown) by addn. of *N* HCl in Et<sub>2</sub>O or 5*N* aq. HCl with cooling and sublimation of the evapn. residue: *H*, m. 73-4°

4  
ARNOLD, Zdenek and SORM, Frantisek  
Me, m. 88-9.5° (from C<sub>6</sub>H<sub>6</sub>); Et, m. 69-70°; Pr, m. 58.5°;  
iso-Pr, m. 62-3°; Bu, m. 54-5°; C<sub>6</sub>H<sub>5</sub>, m. 55°; Ph, m.  
92-3°; P<sub>2</sub>CH<sub>5</sub>, m. 136-7° (from (CH<sub>2</sub>Cl)<sub>2</sub>). Reaction  
mechanisms are discussed. L. J. Urbánek

3/3

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CZECHOSLOVAKIA / Organic Chemistry. Synthetic  
Organic Chemistry.

G

Abs Jour : Ref. Zhur. - Khimiya, No. 15, 1958, No. 50339

Author : Ratusky, Josef; Sorm, Frantisek

Inst : -

Title : Reactions of Diazoketones. II Reactions of  $\omega$ -  
diazooacid Esters With Indole and N-methyl-  
pyrrole.

Orig Pub : Chem. Listy, 1957, 51, #6, 1091-1100.

Abstract : Ethyl (Ia) and methyl (Ib) diazopyroacetate, as  
well as ethyl  $\gamma$ -diazooacetoacetate (II) and  
 $\delta$ -diazolevullinate (III) were synthesized.  
The reactions of these compounds with indole  
(IV) and with methyl pyrrole (V), in presence  
of Cu, were studied. In both cases substitution  
of H by an ester radical takes place with elimi-

Card 1/7

CZECHOSLOVAKIA / Organic Chemistry. Synthetic  
Organic Chemistry.

G

Abs Jour : Ref. Zhur. - Khimiya, No. 15, 1958, No. 50339

nation of  $N_2$ , examples in (IV) in the 3rd posi-  
tion, and in (V) in the 2nd position. The  
structure of diazo compounds (DC) was proven by  
reactions with hydrogen halides,  $CH_3COOH$  and by  
Wolf's rearrangement. DC were synthesized by  
stirring an ether solution of  $CH_2N_2$  with a corres-  
ponding ester chloride for 30 min., at  $-5$  to  $-8^\circ$   
Listed below are: % yield, M.P. in  $^\circ C$ , or B.P.  
 $^\circ C/mm$  and derivative: Ia, 90.2, 73-74,  
ethoxyallyl chloride; Ib, 90, 103-105,  
methoxyallyl chloride; II, 77, 70/0.05,  $C_2H_5OCOCH_2-$   
 $COCl$ ; III 82, 75/0.01,  $C_2H_5OCOCH_2CH_2COCl$ . Reac-  
tion with IV was accomplished by introducing DC  
into a solution of IV in the presence of Cu ( $85-$   
 $95^\circ C$ , 5-10 min.). From a solution of I in

Card 2/7

CZECHOSLOVAKIA / Organic Chemistry. Synthetic  
Organic Chemistry.

G

Abs Jour : Ref. Zhur. - Khimiya, No. 15, 1958, No. 50339

over  $Al_2O_3$ . Upon boiling of the latter (1 hour) with a 30% KOH 3-indolyl acetate, M.P.  $165^{\circ}$  was formed. III in cyclohexane led to ethyl  $\delta$ -3 (indolyl)-levullinate, 85% yield. Oxidation with  $O_2$  yielded indole carboxylic-3 acid, detected by chromatography on paper. Reactions of V are analogous, but take from 0.5 - 2 hours (the extent of reaction being determined by the quantity of liberated  $N_2$ ). I without a solvent yielded ethyl 2- (N-methyl pyroyl) pyroacetoacetate, 39% yield B.P.  $110^{\circ}/0.25$ mm which yielded with a 3N NaOH 2-(N-methylpyroyl) pyroacetate, M.P.  $112^{\circ}$ . From II in cyclohexane may be obtained ethyl  $\gamma$ -(2-N-methyl pyroyl)-acetoacetate, 23.2% yield, B.P.  $90^{\circ}/0.5$  mm. Heating of the latter for 15

Card 4/7

CZECHOSLOVAKIA / Organic Chemistry. Synthetic  
Organic Chemistry.

G

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min. resulted in the formation of 2-N-methyl-pyroylacetone, (semicarbazone), M.P.  $166^{\circ}$ . III in  $C_6H_6$  yielded ethyl  $\delta$ -2-N-methylpyroyl levullinate, 65.8% yield, B.P.  $114^{\circ}/0.2$  mm. In order to prove the structure of DC the following compounds were synthesized: methyl ethyl chloropyroacetate from Ib and HCl (g) in  $CH_3OH$ , 94.5% yield, B. P.  $78^{\circ}/13$  mm which with a 35% HCl (acid) gave chloropyroacetic acid, M.P.  $57^{\circ}$ ; ethyl bromopyroacetate, B.P.  $87^{\circ}/9$  mm, obtained by shaking Ia in ether with a 40% HBr. At the same time bromopyroacetic acid, M.P.  $75^{\circ}$  was formed. By analogy, II and HCl (g) yielded ethyl  $\gamma$ -chloroacetoacetate, 66.5% yield, B.P.  $95^{\circ}/10$  mm,  $n_D^{20}$  1,4550 (tear gas), while II

Card 5/7

CZECHOSLOVAKIA / Organic Chemistry. Synthetic  
Organic Chemistry.

G

CZECHOSLOVAKIA / Organic Chemistry. Synthetic  
Organic Chemistry.

G

Abstr Jour : Ref. Zhur. - Khimiya, No. 15, 1958, No. 50339

$\zeta$ -acetoxyacetoacetate 86% yield, B.P. 70°/0.1 mm, and ethyl -acetoxylevullinate, 84% yield B.P. 71°/0.0005 mm. Heating of alcoholic solutions of DC with AgNO<sub>3</sub> yielded complex esters. Listed below are esters and % (yields): CH<sub>2</sub> (COOCH<sub>3</sub>)<sub>2</sub> (from Ib), 66.5; ethyl succinate (from II), 67; ethyl glutarate (from III), 90.5; Communication I see Ref. Zhur. Khim., 1958, 14358.  
-- T. Kovaz

Card 7/7

Proteins. XL. Comparison of weakly basic peptides  
 from partial hydrolyzates of chymotrypsinogen and trypsin.  
 Jiří Vaněček, Bedřich Mělník, and František Šerpi (Czech-  
 oslov. akad. věd, Prague). *Chem. listy* 51, 1367-75 (1957);  
 cf. *C. A.* 51, 12188e. — Partial hydrolysis of chymotrypsino-  
 gen (I) and trypsin diisopropyl phosphate (II) (carried out  
 by treating 5 g. of the protein with 150 ml. concd. HCl at  
 37° 8 days, dilg. with H<sub>2</sub>O to 130 ml., evapg. to dryness *in*  
*vacuo* at 35°, and removing the residual HCl by repeated  
 evapn. with H<sub>2</sub>O), and combined electrophoresis at pH 7  
 in a 5-chamber app., descending paper electrophoresis at  
 pH 5.6, and paper chromatography yielded a series of pep-  
 tides, most of which are common both for I and II. The  
 following peptides were identified from I and II, resp.:  
 alanylhistidine (I, II), histidyl-3-sulfoalanine (I, II), phen-  
 ylalanylhistidine (I), histidylphenylalanine (I), tyrosylhis-  
 tidine (II), prolylhistidine (II), lysylglutamic acid (I, II),  
 lysylglutamylisoleucine (I, II), lysylserine (I, II), lysylleu-  
 cine (I, II), lysyl-3-sulfoalanine (I, II), 3-sulfoalanylglu-  
 tamic acid (I, II), 3-sulfoalanylthreonine (I), 3-sulfoalkaryl-  
 valine (I), 3-sulfoalanylserine (II). M. Hudlický

4

*M. Hudlický*

CZECHOSLOVAKIA / Organic Chemistry. Synthetic Organic G-2  
Chemistry.

Abs Jour: Ref Zhur-Khimiya, No 23, 1958, 77633.

Author : Novak, J. and Sorm, F.

Inst : Not given.

Title : Reactions of Diazoketones. III. The Reaction of  
Diazoacetone with Furan and With Its Homologs.  
A New Simple Synthesis of Octa-3,5-diene-2,7-dione.

Orig Pub: Chem Listy, 51, No 9, 1693-1698 (1957) (in Czech).

Abstract: In the course of their investigation of the feasi-  
bility of the utilization of diazoacetone (I) in  
the synthesis of other compounds, the authors have  
reacted I with furan (II) and with its homologs  
using 1 : 1 mol ratios. The reaction is accompan-  
ied by the evolution of N<sub>2</sub> and the opening of the  
ring of II. With II, 2, 4-heptadiene-6-one-al

Card 1/4

26

CZECHOSLOVAKIA / Organic Chemistry. Synthetic Organic G-2  
Chemistry.

Abs Jour: Ref Zhur-Khimiya, No 23, 1958, 77633.

Abstract: of NaBrO in the cold gives trans, trans-muconic acid (VII), yield 70%, mp 303° (from water). VII is also obtained by the direct oxidation of III with NaBrO. The Hydrogenation of III over PtO<sub>2</sub> in alcohol gives 6-heptanone-1-al, bp 98-102°/ 15mm; hydrogenation over Raney Ni (120 atm) gives 1,6-heptanediol, bp 80°/0.3mm. The reaction of I with IIa is carried out by adding the mixture of reactants to a boiling suspension of Cu in IIa; the yield of IV is 39%, mp 127° (from CH<sub>3</sub>OH). V is obtained by a similar procedure from I and IIb, yield 38%, bp 93-96°/0.3mm, mp 38°. The ethyl ester of diazoacetic acid under similar conditions

Card 3/4

27

CZECHOSLOVAKIA/Organic Chemistry. Natural Substances and  
Their Synthetic Analogs.

G

Abs Jour: Ref. Zhur-Khimiya, No 19, 1958, 64585.

(absorption line at  $1737\text{ cm}^{-1}$ ). Oxidation of formyl-acoranone (III) gives dicarboxylic acid (IV). Catalytic reduction of (I) leads to the keto-alcohol (V), produced by the dehydrogenation of acorenone (VI) with CO-groups in the 5-member ring. Upon dehydrogenation with S, (VI) yields a product whose UV-spectrum reveals lines at  $1714\text{ cm}^{-1}$  (CO-group in 6-member ring) and  $1514\text{ cm}^{-1}$  (aromatic system). In this fashion, dehydrogenation is accompanied by the enlargement of the 5-member ring, from which it follows that one of the C atoms, common to the 5- and 6-member rings, appear to be quaternary. Upon dehydrogenation of (IV), there are produced 1-methyl-4-ethylbenzol (VII), 1-methyl-4-isobutylbenzol (VIII), propionic acid (IX), and isovaleric acid (X).

Card : 2/10

36

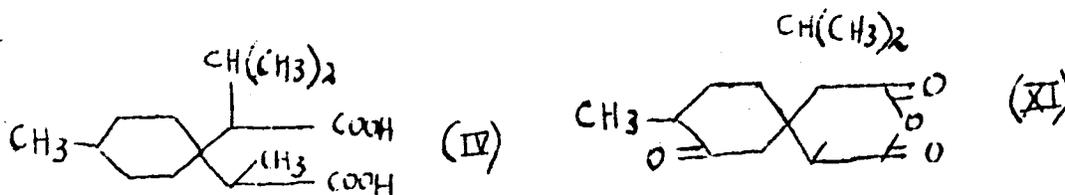
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CIA-RDP86-00513R001652420016-4"

CZECHOSLOVAKIA/Organic Chemistry. Natural Substances and  
Their Synthetic Analogs.

G

Abs Jour: Ref. Zhur-Khimiya, No 19, 1958, 64585.



Monobenzilidene-derived (I) yields, upon ozonification, the keto-anhydride (XI). Upon pyrolysis of the Ba salts of dicarboxylic acid (XII), derived from (XI), there are separated (IX), (X) and two ketones,  $C_9H_{14}O$  (XIII) and  $C_{11}H_{20}O$  (XIV), whose carbon skeleton has

Card : 3/10

CZECHOSLOVAKIA/Organic Chemistry. Natural Substances and  
Their Synthetic Analogs.

G

Abs Jour: Ref. Zhur-Khimiya, No 19, 1958, 64585.

Herout, Collect. czechoslow. chem. Commun., 1949, 14, 723) give, upon dehydration with phthallic anhydride (260-280°, 1 hour) (VI), yield 2.84 g. b.p. 147-151°/13 mm. Dehydrogenation of 1.65 g. of (VI) with S (0.63 g) (3 hours, 220-230°), and chromatographing the product on Al<sub>2</sub>O<sub>3</sub>, gave 0.55 g. of a substance (b.p. 140-160°) temp. bath/13 mm. From (I) (2.7 g) by means of ethanedithiol (2.65 g) and ethered BF<sub>3</sub> (3 ml), there was produced the monothioethyleneketal of (I) (Ia), m.p. 77-78.5°,  $[\alpha]_D^{20} + 111.6^\circ$  (w. 4.42); from the mother solution, by way of chromatographing on Al<sub>2</sub>O<sub>3</sub> (washed with C<sub>6</sub>H<sub>6</sub>) there was separated. the bithioethyleneketal of (I), m.p. 95-97° (in petr. ether)  $[\alpha]_D^{20} + 0$  (w. 2.3).

Card : 5/10

CZECHOSLOVAKIA/Organic Chemistry. Natural Substances and  
Their Synthetic Analogs.

G

APPROVED FOR RELEASE: 08/25/2000 CIA-RDP86-00513R001652420016-4"

Abs Jour: Ref. Zhur-Khimiya, No 19, 1958, 64585.

From 4.6 g. of (Ia), boiled with elemental Ni in dioxane (10 hours), (II) was produced, yield 2.9 g., b.p. 151-154°/15 mm,  $[\alpha]_D^{20} + 47.0^\circ$  (w. 2.2, chloroform,  $n_D^{20} 1.4857$ ,  $d_4^{20} 0.9539$ . By the action of CH<sub>3</sub>ONa (from 1.4 g. Na) and HCOC<sub>2</sub>H<sub>5</sub> on (II) (2.7) in C<sub>6</sub>H<sub>6</sub> (20 ml) (20°, 40 hours, atmosphere N<sub>2</sub>), (V) is synthesized, yield 2.6 g., b.p. 110-114°/0.04 mm, 142-147°/2.5 mm,  $[\alpha]_D^{20} + 29.6^\circ$  (w 3.17). To a solution of (III) (5.7 g) and KOH (11.4 g) in CH<sub>3</sub>OH (225 ml) is added 30% H<sub>2</sub>O<sub>2</sub> (225 ml); after boiling for 50 minutes, 5.2 g of (IV) are precipitated (not crystalline); dimethylether (IV), b.p. 120-126°/0.05 mm. Upon dehydrogenation with 5% Pd/C (310-330°, 3-9 hours), (IV) gives a mixture of (VII) and (VIII), b.p. 70-95° (temp. bath)/

Card : 6/10

38

CZECHOSLOVAKIA/Organic Chemistry. Natural Substances and  
Their Synthetic Analogs.

G

Abs Jour: Ref. Zhur-Khimiya, No 19, 1958, 64585.

over 5%; Pd/SrCO<sub>3</sub> (1.5 g.) and distillation in a column yields (XIII) (4.3 g), b.p. 73-78°/10mm; semicarbazone, m.p. 152-154° (in 70% al.); (XIV), yield 2.7 g., b.p. 97-102°/10 mm; semicarbazone, m.p. 172° (in 90% al.). (XIII) (1.03 g) is reduced by LiAlH<sub>4</sub> (0.55 g) in ether (20 ml) (boiled for 10 minutes), the products of the reaction (b.p. 93-95°/14 mm., 0.94 g.) being dehydrogenated by heating with phthalic anhydride (40 minutes, 270-280°). Chromatographing on Al<sub>2</sub>O<sub>3</sub>, a product is separated (b.p. 80-90°, (bath temp.)/62 mm, 0.57 g) which can be dehydrogenated with 0.37 g of S (8 hours, 220-250°) and chromatographed on Al<sub>2</sub>O<sub>3</sub> to produce (VII) yield 0.17 g., b.p. 70-80° (bath temp.)/35 mm, n<sub>D</sub><sup>20</sup> 1.4953, d 0.8585. Similarly, (XIV) (1 g) can be reduced to a product

Card : 8/10

39

APPROVED FOR RELEASE: 08/25/2000

CIA-RDP86-00513R001652420016-4"

CZECHOSLOVAKIA/Organic Chemistry. Natural Substances and  
Their Synthetic Analogs.

G

Abs Jour: Ref. Zhur-Khimiya, No 19, 1958, 64585.

with a b.p. of 108-113°/13 mm (0.95 g), which upon dehydration gives a product with b.p. 107-117° (bath temp) 57 mm (0.62 g); dehydrogenation of this product gives (VIII), b.p. 80-90° (bath temp.)/13 mm, 0.17, n<sub>D</sub><sup>20</sup> 1.4880, d<sub>4</sub><sup>20</sup> 0.8523. By comparison, (X) is synthesized out of 4-methylcyclohexanone (2.2 g) boiled with isobutyl lithium (out of 1.9 g Li and 18 ml isobutyl chloride) in petr. ether for 1 hour, giving 1-methyl-4-isobutylcyclohexanone-4, yield 2.85 g, b.p. 90-101°/11 mm. Dehydration of 2.62 g. of this latter with 85% HCOOH (100°, 1 hour) gives 2.05 g of a hydrocarbon with b.p. 64-69°/12 mm., the dehydrogenation of which (1 g) with the aid of 0.47 g S (200-250°, 8 hours), produces (VIII). The reduction of 20 g (XV) with LiAlH<sub>4</sub>

Card : 9/10

CZECHOSLOVAKIA/Organic Chemistry. Natural Substances and  
Their Synthetic Analogs.

G

Abs Jour: Ref. Zhur-Khimiya, No 19, 1958, 64586.

benzyl alcohol there is derived a product (350 mg),  
 $n_D^{20}$  1.5030,  $d_4^{20}$  0.9505. The latter by being  
hydrolyzed in alcohol in the presence of an alcohol  
solution of KOH and 5% Pd/SrCO<sub>3</sub> (1 g), distilled  
(118-126°/20 mm), dehydrogenated with 50 mg [omission  
transl] (180-220°, 4 hours), chromatographed on Al<sub>2</sub>O<sub>3</sub>  
in gasoline and cleaned by picrate, m.p. 92° (in alc.)  
this product is converted into 1,7-dimethyl-4-isopropyl-  
naphthalene (120 mg), m.p. 60° (in alc.); styphnate, m.p.  
124° (in alc.). Oxidizing (II) (2 g) with HIO<sub>4</sub>·2H<sub>2</sub>O (4.4 g)  
in 0.5 l water (20°, 2 days) produced HCOOH. The saponi-  
fication of (I) (2 g) with a 5% methanol solution of KOH  
(20°, 1 hour) produced laserol (III), yield 79%, m.p. 190°  
(in ethyl acetate). The oxidation of (III) (1 g) with

Card : 2/7

4/

"APPROVED FOR RELEASE: 08/25/2000

CIA-RDP86-00513R001652420016-4"

CZECHOSLOVAKIA/Organic Chemistry. Natural Substances and  
Their Synthetic Analogs.

G

Abs Jour: Ref. Zhur-Khimiya, No 19, 1958, 64586.

CrO<sub>3</sub> in pyridine (20°, 24 hours) produced laserone,  
m.p. 180-181° (in ethylacetate). From 0.5 g (III), by  
the action of ClCOOC<sub>2</sub>H<sub>5</sub> (1 ml) in dioxane (5 ml)  
and pyridine (0.8 ml) (20°, 4 hours), and by boiling  
for 30 minutes in 2% HCl (13 ml), there was derived  
mono-O-carboethoxyl laserol, m.p. 144°, (in acetone).  
This latter, (0.7 g) was oxidized with CrO<sub>3</sub> (1 g) in  
pyridine (20 ml) (30°, 24 hours) to mono-O-carbethoxy-  
diketone Cl<sub>5</sub>H<sub>12</sub>O<sub>7</sub>, m.p. 98-99° (in ether-gasoline). The  
dehydration of (I) (1 g) with SOCl<sub>2</sub> (12 ml) in pyridine  
(15°, 10 minutes) gave unsaturated hydroxyketodiether  
C<sub>25</sub>H<sub>36</sub>O<sub>6</sub> (IV), yield 0.7 g, m.p. 110°. Hydrogenation  
of (IV) (169 mg) over Pt (from PtO<sub>2</sub>) in glacial CH<sub>3</sub>COOH,  
and chromatographing the products of the reaction on Al<sub>2</sub>O<sub>3</sub>

Card : 3/7

CZECHOSLOVAKIA/Organic Chemistry. Natural Substances and  
Their Synthetic Analogs.

G

Abs Jour: Ref. Zhur-Khimiya, No 19, 1958, 64586.

on nitr.  $Al_2O_3$  in gasoline gave a dimethyl ether ketodicarboxylic acid  $C_{17}H_{32}O_5$ , yield 150 mg. Dehydration of tetrahydroxylaserpitine (XI) (2g) (see reference above) with  $SOCl_2$  (10 ml) in pyridine (30 ml) at  $15^\circ$  and chromatographing the results of the reaction on nitr.  $Al_2O_3$  gave an unsaturated hydroxyketodiether  $C_{25}H_{40}O_6$  (XII), b.p.  $134-142^\circ/0.3$  mm, yield 1.8 g. This latter can be reduced with  $LiAlH_4$  (like V) to give VII, m.p.  $183-184^\circ$ . Dehydration of (XI) (2.5 g) with  $SOCl_2$  (5 ml) at  $80^\circ$  in pyridine yields an unsaturated ketodiether  $C_{25}H_{38}O_5$ , yield 1.7 g., m.p.  $75^\circ$  (in alc.). This latter, upon reduction with  $LiAlH_4$ , gives (IX), yield 0.65 g. (XI) could not be oxidized by

Card : 6/7

43

CZECHOSLOVAKIA/Organic Chemistry. Natural Substances and  
Their Synthetic Analogs.

G

Abs Jour: Ref. Zhur-Khimiya, No 19, 1958, 64586.

$CrO_3$  in glacial  $CH_3COOH$  or pyridine. Added are the UV-spectrum data of (III), (IV), (VIII) and (XII).

Card : 7/7

CZECHOSLOVAKIA/Organic Chemistry. Natural Products and Their  
Synthetic Analogues.

G-3

Abs Jour: Ref Zhur-Khim., No 24, 1958, 81783

Author : Cerny V., Labler L., Sorn F.  
Inst :  
Title : The Steroids XXXIV. The Structure of Holarrhedine

Orig Pub: Chem. listy, 1957, 51, No 12, 2351-2355.

Abstract: The authors have suggested a structural formula  
3 $\alpha$ , 20 $\alpha$ -diamino-18-oxy- $\Delta^5$ -pregnen for  
holarrhydine (I) based on the results of thermal  
splitting and leading to the products which were  
identical with the products of the splitting of  
holarrhymine (II), based on the difference of  
corresponding desoxitetramethyl derivatives of I

Card : 1/4

CZECHOSLOVAKIA/Organic Chemistry. Natural Products and Their  
Synthetic Analogues

0-3

Abs Jour: Ref Zhur-Khim., No 24, 1958, 81783.

and II, and on the difference of the molecular ro-  
tation between I and II, which corresponds to the  
difference between 3 $\alpha$ -amino- $\Delta^5$ -steroid and  
3 $\alpha$ -epimer. N, N, N', N'-tetramethyl holarrhymine  
(III) was allowed to stand in aqueous acetic acid  
and CrCl<sub>3</sub> for 12 hours, and 3 $\beta$ , 20 $\alpha$ -bis-  
dimethylamino- $\Delta^5$ -pregnenal-18, C<sub>27</sub>H<sub>42</sub>ON<sub>2</sub> sepa-  
rated, m.p. 141-143°C.,  $[\alpha]_D^{20} + 8^\circ$ , which by heating  
for 3 hours with N<sub>2</sub>H<sub>4</sub>·H<sub>2</sub>O and KOH in triethylene  
glycol to 200-215°C. produced 3 $\alpha$ , 20 $\alpha$ -bis-  
dimethylamino- $\Delta^5$ -pregnen, m.p. 140-141°C.,  
 $[\alpha]_D^{20} - 40^\circ$ . In the same way from N, N, N', N'-te-  
tramethyl-I (IV) was obtained 3 $\alpha$ , 20 $\alpha$ -bis-dimethylamino-  
 $\Delta^5$ -pregnenal-18, m.p. 164-165°C.,  $[\alpha]_D^{20} + 12^\circ$ ,

Card : 2/4

Diatri 4E2a(1)

A contribution to the structure of absinthin and anabsinthin / L. Novotný, V. Herout, and E. Šorm (Czechoslovak Acad. Sci., Prague). *Chem. & Ind. (London)* 1958, 485-6.

The mol. wt. of cryst. anhyd. absinthin (I), m. 179-80° (decompn.), detd. by the isothermal distn. method in Me<sub>2</sub>CO, is 520 ± 50, thus I is a dimeric guanoinkide, C<sub>20</sub>H<sub>28</sub>O<sub>4</sub>, contg. 2 OH groups/mol. Anabsinthin (II) has the same mol. wt. and mol. formula. The infrared spectrum of I shows absorption bands due to a γ-lactone carbonyl group (1768 cm.<sup>-1</sup>), a double bond (1662 cm.<sup>-1</sup>), and an OH group (3530 cm.<sup>-1</sup>). Hydrogenation of I with PtO<sub>2</sub>-HOAc did not proceed readily, but gave tetrahydro-I (III), C<sub>20</sub>H<sub>30</sub>O<sub>4</sub>, m. 282°, and II, proving the presence of 2 double bonds in I. The OH groups in III are tertiary, since they could be neither oxidized with CrO<sub>3</sub> nor acetylated. Reduction of I with LiAlH<sub>4</sub> gave I diastol (IV), C<sub>20</sub>H<sub>30</sub>O<sub>4</sub>, m. 195°; use of LiAlH<sub>4</sub> in N-ethylpiperidine gave a glassy product, probably a tetrol (V), formed by subsequent etherification of 2 OH groups. Dehydrogenation of IV with Se gave chamazulene, guaiazulene, and artemizulene (VI) (trinitrobenzene adduct, m. 191°); similar dehydrogenation of V gave the above azulenes and 12-hydroxyguaiazulene (trinitrobenzene adduct,

Y. Oc.

m. 148°) as the principal product. Infrared bands due to double bonds and OH groups are absent in the spectrum of II, which does not take up H even under forcing conditions. Reduction of II with LiAlH<sub>4</sub> in Et<sub>2</sub>O gave a product, C<sub>20</sub>H<sub>30</sub>O<sub>4</sub>, m. 211°, probably a pentol, which on dehydrogenation with Se gave VI in fair yield. Oxidation of I and II with concd. HNO<sub>3</sub> gave a mixt. which chromatographed on

Dowex-2 gave HO<sub>2</sub>CCH<sub>2</sub>CH(CHMe)CO<sub>2</sub>CHCO<sub>2</sub>H (VII), m. about 165°, and liquid HO<sub>2</sub>CCH<sub>2</sub>CH(CO<sub>2</sub>H)CHMeCO<sub>2</sub>H (VIII), identified as its tri-Me ester. The infrared spectrum of the di-Me ester of VII has bands at 1744 cm.<sup>-1</sup> (ester group) and 1792 cm.<sup>-1</sup> (γ-lactone). VII and VIII also were obtained from artabsin (IX) by a similar procedure. Treatment of both I and II in alk. soln. with HCHO gave

1,3,4-trimethyl-7-ethylazulene and with AcH gave 1,4-dimethyl-3,7-diethylazulene. It is assumed that I is formed by diene addn. of an unknown precursor contg. a cyclopentadiene group and an OH group in the 10-position of guaiazulide (isomer of IX). The structure of I is proposed as shown, which permits the stereochem. formation of II from I by the probable addn. of 1 of the OH groups to the double bond, affording the oxide ring.

Rip G. Rice

6  
2-May  
1

KEILOVA, Helena, (Chemicky ustav CSAV, Praha 19, Na cvicisti 2.) SORM, Frantisek,  
(Chemicky ustav CSAV, Praha 19, Na cvicisti 2.)

A contribution to the study of resistance to Crocker's tumour. Neoplasma,  
Bratisl. 5 no.1:26-33 1958.

1. Institute of Chemistry, Czechoslovak Academy of Sciences, Prague.  
(NEOPLASMS, exper.  
Crocker's tumor, ligation & transpl. time in relation to  
immunity)

WINKLER, Alojz; UJHAZY, Viliam; KNOTZOVA, Kornelia; SORM, ~~Trantisek~~

The level of 6-azouracil in the serum of rats. Neoplasms, Bratisl.  
5 no.2:97-100 1958.

1. Oncological Research Institute, Bratislava. Chemical Institute of  
the Czechoslovak Academy of Sciences, Praha. Authors' address: Dr. A.  
Winkler and co-workers, Bratislava, ul. Cs. armady 17; akademik F. Sorm,  
Praha, Chemicky ustav CXAV.

(URACIL, antag.

6-azouracil, retention in blood in rabbits)

(CYTOTOXIC DRUGS, in blood,

6-azouracil, retention in rabbits)

EXPERIENTIA MEDICA Sec 2 Vol 12/5 Physiology May 59

2022. THE ANTI-TUMOUR ACTIVITY OF 6-AZAZURACIL RIBOSIDE - Šorm  
F. and Keilová H. Inst. of Chem., Czechoslovak Acad. of Sci., Prague-  
EXPERIENTIA (Basel) 1958, 14/6 (215) Tables 1  
6-Azauracil riboside is 6 times as active as 6-azauracil against the s. c. form of  
Ehrlich ascites tumour. Fischer - Buenos Aires (11, 5, 16)

Country : CZECHOSLOVAKIA G  
Category : Organic Chemistry. Natural Substances and  
Their Synthetic Analogs  
Abs Jour : Ref Zhur - Khim., No 5, 1959, No. 15499  
Author : Sykora, V.; Herout, V.; Pliva, J.; Sorm, F.  
Inst : -  
Title : Terpenes. LXXXII. Structure of Acoron  
Orig Pub : Collect. czechosl. chem. commun., 1958, 23,  
No 6, 1072-1082  
Abstract : No abstract.  
See Ref Zhur-Khim, 1958, 64585.

Card: 1/1

G - 78

*Sorm, F.*  
APPROVED FOR RELEASE: 08/25/2000 CIA-RDP86-00513R001652420016-4"

CZECHOSLOVAKIA/Organic Chemistry. Organic Synthesis. G-2

Abs Jour : Ref Zhur-Khimiya, No 9, 1959, 31331

Author : Novak, J., Sorm, F.

Inst : -

Title : Reactions of Diazoketones. III. Interac-  
tion of Diazoacetone with Furan and Its  
Homologues. New Simple Synthesis of Octa-  
-3,5-Dienedione-2.7.

Orig Pub : Collect. czechosl. chem. commun., 1958,  
23, No 6, 1126-1132

Abstract : See Ref Zhur-Khimiya, 1958, 77633.

Card : 1/1

138

SORM, F

CZECHOSLOVAKIA / Organic Chemistry--Natural compounds 0-3  
and their synthetic analogs

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27578

Author : Motl, O., Herout, V., and Sorm, F.; Motl, O.,  
Sykora, V., Herout, V., and Sorm, F.

Inst : Not given

Title : On Terpenes. LXXXV. The Structure of 'Juniper  
Camphor', Another Alcohol of the Selinane-Type  
from Juniper Oil. LXXXVI. The Structure of Two  
Crystalline Cadinols.

Orig Pub: Collection Czechoslov Chem Commun, 23, No 7,  
1293-1296 and 1297-1306 (1958) (in English with  
a Russian summary)

Abstract: See RZhKhim, 1959, 8394, 8395.

Card 1/1

117

SOHN, P.; KRYANEK, M.; JERUSA, V.

"Terpenes. XCIII. Composition of essent oil (from Saussurea lappa Clarke)."  
(In English)

COLLECTION OF CZECHOSLOVAK CHEMICAL COMMUNICATIONS, Praha, Czechoslovakia,  
Vol. 23, no. 12, Dec. 1958

Monthly list of EAST EUROPEAN ACCESSIONS, (EMEA), IC, Vol. 7, No. 7, July 1959, Unclas.

SMIL, P.; NICKY, M.

"Terpenes. XC. Synthesis of 1, 7-dimethyl-4-isopropylcyclodecane (germacrane).  
(In English)

COLLECTED CHEMICAL COMMUNICATIONS, Praha, Czechoslovakia,  
Vol. 23, no. 12, Dec. 1959

Monthly List of EAST EUROPEAN JOURNAL OF CHEMISTRY (EJEC), IC, Vol. 8, No. 7, July 1959, Unclas.

RYN, E.; FABRYAN, V.

"Mechanism of the antitumor action of 6-azauracil and its riboside."  
(In English)

CONTRIBUTION OF A SCIENTIFIC JOURNAL COMMUNICATIONS, Praha, Czechoslovakia,  
Vol. 23, no. 12, Dec. 1959

Monthly list of EAST EUROPEAN ACCESSIONS (EEAI), IC, Pol. 8, No. 7, July 1959, Unclas.

SORM, F.

CZECHOSLOVAKIA / Organic Chemistry. Natural Compounds G-3  
and Their Synthetic Analogues.

Abs Jour: Ref Zhur-Khimiya, No 3, 1959, 8394.

Author : Motl, Otakar., Herout, Vlastislav, Sorm, Frant-  
isek.

Inst : Not given.

Title : On Terpenes. LXXV. Structure of "Juniper Cam-  
phor", a New Alcohol of Selinán-Type, from Essen-  
tial Oil of Juniper.

Orig Pub: Chem. listy, 1958, 52, No 1, 116-119.

Abstract: It is demonstrated that the so-called "juniper  
camphor" isolated from *Juniperus Communis* L. is  
1.10-dimethyl-7-isopropylidene-decalol-1(Δ<sup>7</sup>(11)-  
selinanol-4) (I). Hydrogenation of I gives sel-  
icanol-4 (II), which on dehydration with SOCl<sub>2</sub>  
in pyridine is converted to a mixture of hydro-

Card 1/3

CZECHOSLOVAKIA / Organic Chemistry. Natural Compounds G-3  
and Their Synthetic Analogues.

abs Jour: Ref Zaur-Khimiya, No 3, 1959, 8395.

Author : Motl, Otakar, Sykora, Vladimir., Hecrut Vlast-  
inil., Sora Frantisek.

Inst : Not given.

Title : On Terpenes. LXXXVI. On the Structure of Two  
Crystalline Cadinols.

Orig Pub: Chem. listy, 1958, 52, No 2, 316-324.

Abstract: It was ascertained (on the basis of comparison  
of infrared spectra and physical constants) that  
two alcohols isolated from essential oil of jun-  
iper, are  $\Delta^1$ -cadinol-10 (I) and  $\Delta^1$ (10)-cadi-  
nol-4 (II). I is identical with the cadinol  
previously isolated (Platter P.A., Markus R.,  
Helv. chim. acta, 1942, 25, 1674) from essential

Card 1/9

97

CZECHOSLOVAKIA / Organic Chemistry. Natural Compounds G-3  
and Their Synthetic Analogues.

Abs Jour: Ref Zhur-khimiya, No 3, 1959, 6395.

Abstract: hydrocarbon mixture gave 4-isopropyl-6-methyl-decalone (IX). On the basis of the course of these reactions it can be considered that the hydroxy-group of I is equatorial, which is in accord with the ready ester-formation by the hydroxyl group. On dehydration-dehydrogenation with S, II is converted to cadalin. Hydrogenation of II gave cadinol-4 (X), which yielded on dehydration a hydrocarbon from which was synthesized, by ozonization, a diketo-alcohol (XI). Under ordinary conditions II is not esterified, which indicates an axial position of OH. Since relative configuration of cadinene dihydrobromide had been determined previously, (see RZhKhim, 1958, 76438), the authors attribute to I and II the

Card 3/9

98

CZECHOSLOVAKIA / Organic Chemistry. Natural Compounds G-3  
and Their Synthetic Analogues.

Abs Jour: Ref Zhur-Khimiya, No 3, 1959, 8395.

Abstract: Slovak origin, by fractionation of sesquiterpene alcohols in a column having an efficiency of 30 theoretical plates, followed by crystallization of the fraction of BP 123-125°/2mm. I, MP 74.5°,  $[\alpha]_D^{20}$  - 47° (c 2.00; chloroform), was obtained, a) from essential oil of juniper mother-liquor (following separation of II), over the p-nitrobenzoate, BP 138°,  $[\alpha]_D^{20}$  - 67.4°; b) from essential oil of citronella, by distillation in a glass bubbler-cap column (34 theoretical plates) from the fraction of BP 134-139°/2 mm, after chromatography on Al<sub>2</sub>O<sub>3</sub> (activity III-IV) and purification over the p-nitrobenzoate. Into a solution of 311 mg I in a mixture of 20 ml dioxane and 5 ml water.

Card 5/9

99

CZECHOSLOVAKIA / Organic Chemistry. Natural Compounds G-3  
and Their Synthetic Analogues.

Abs Jour: Ref Zhur-khimiya, No 3, 1959, 6395.

Abstract: -cadinandiol-8,10, yield 0.4 g, MP 209-210°. De-  
hydration of the latter (0.3 g) by heating with  
4 ml CH<sub>3</sub>COOH and 2 drops 70% HClO<sub>4</sub> (30 minutes  
at 100°), and dehydrogenation of the product with  
0.11 g S (3 hours at 210-250°) followed by chrom-  
atography on alkaline Al<sub>2</sub>O<sub>3</sub>, give cadalin; pi-  
crate, MP 113°. Reduction of 1.5 g Vb with  
LiAlH<sub>4</sub> gives VI, yield 1.3 g, MP 214-216°. Ox-  
idation of 0.85 g VI with CrO<sub>3</sub> in CH<sub>3</sub>COOH gives  
VII, MP 109-109.5° (from petroleum ether). Re-  
action of 430 mg VII with CH<sub>3</sub>I in ether, de-  
hydration of the product (0.44 g) in CH<sub>3</sub>COOH  
4.4 ml with 2 drops of HClO<sub>4</sub> (1 hour, 100°), and  
chromatography on Al<sub>2</sub>O<sub>3</sub>, followed by dehydrogen-

Card 7/9

100

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds and Their Synthetic Analogs.

G

Abstr Jour: Ref Zhur-Khin., No 11, 1959, 3877<sup>4</sup>.

Author : Holub, M., Herout, V., and Sorn, F.

Inst :  
Title : Chemistry of the Terpenes. LXXXVII. Proof of the Structure of Laserpitine.

Orig Pub: Chem Listy, 52, No 3, 498-504 (1958) (in Czech);  
Collection Czechoslov Chem Commun, 23, No 7, 1307-1313 (1958) (in English with a Russian summary);

Abstract: The formula proposed earlier by the authors for Laserpitine I,  $R = CH_2CH=C(CH_3)CO$  (Kuz'min, 1958, 64586) is confirmed by the proof of the structure of dihydrolaserol [spelling uncertain] (II) and of the triol (III) [see insert below]. The ox-

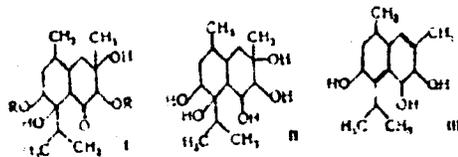
Card : 1/6

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds and Their Synthetic Analogs.

G

Abstr Jour: Ref Zhur-Khin., No 11, 1959, 3877<sup>4</sup>.

oxidation of II by  $KMnO_4$  gives a mixture of acids from which  $\beta$ ,  $\xi$ -dimethyl- $\epsilon$ -keto- $\gamma$ -carboxycaprylic acid (IV), 1-isopropyl-3-methyl-1-hydroxycyclopentane-2,4-dicarboxylic acid (V), methyl succinic acid (VI), betamethylglutaric acid (VII), and  $\beta$ -methyl- $\beta'$ -carboxyglutaric acid (VIII) have been separated. The oxidation of III by



Card : 2/6

G-44

CZECHOSLOVAKIA/ Organic Chemistry. Natural Compounds and Their  
Synthetic Analogs. G

Obs Jour: Ref Zhur-Khin., No 11, 1959, 38774.

$\text{KMnO}_4$  at  $10^\circ$  gives  $\text{CH}_3\text{COOH}$ ,  $(\text{CH}_3)_2\text{CHCOOH}$ , VI, and  
beta-methyl-alpha-carboxylglutaric acid (IX). When  
III was oxidized at  $60^\circ$  only VI could be identified  
among the reaction products. An emulsion of 10 gms  
II in 250 ml 5%  $\text{H}_2\text{SO}_4$  is treated with  $\text{KMnO}_4$  at  $6^\circ$   
until the color no longer disappears, the  $\text{MnO}_2$  formed  
is decomposed by passing  $\text{SO}_2$  into the solution, and  
the latter is extracted with ether. The volatile  
components are removed by steam distillation, the  
neutral substances are separated by extraction with  
ether, leaving 1.25 gms of nonvolatile acids (fraction  
a). On further extraction [ether] an additional  
0.3 gms of nonvolatile acids is obtained (fraction b).

Card : 3/6

CZECHOSLOVAKI./Organic Chemistry. Natural Compounds and Their Synthetic Analogs.

G

Abs Jour: Ref Zhur-Khin., No 11, 1959, 38774.

CH<sub>3</sub>COOH). The triethyl ester of IX, bp 100-103°/1.5-1.7 mm, n<sub>D</sub><sup>20</sup> 1.4384, has been synthesized for comparison purposes by condensing 16 gms CH<sub>2</sub>-(COOC<sub>2</sub>H<sub>5</sub>)<sub>2</sub> and 11.4 gms CH<sub>2</sub>CH=CHCOOC<sub>2</sub>H<sub>5</sub> with C<sub>2</sub>H<sub>5</sub>ONa; saponification of the condensation product (refluxing with 5% alcoholic KOH) yields IX, mp 131-132°. The condensation of 15.7 gms of the triethyl ester of beta-methyl-alpha-carboxyglutaric acid with 20.0 gms BrCH<sub>2</sub>COOC<sub>2</sub>H<sub>5</sub> in the presence of Na in abs ether gives the tetraethyl ester of β-methyl-β', β', -dicarboxydimelic acid (XI acid), yield 62.5%, bp 153-155°/0.3 mm, n<sub>D</sub><sup>20</sup> 1.4490, d<sub>4</sub><sup>20</sup> 1.1090, [R]<sub>D</sub><sup>20</sup> [sic] 87.2 (calculated 86.9); saponification of the latter product gives

Card : 5/6

Synthetic Analogs.

Abs Jour: Ref Zhur-Khin., No 11, 1959, 38774.

APPROVED FOR RELEASE: 08/25/2000 CIA-RDP86-00513R001652420016-4"

XI, mp 150-151° (from conc HCl). When XI is heated for 20 min at 145°/20 mm, VIII is obtained, mp 153-154° (from conc HCl). Data on the UV spectra of IVa and X and on the IR spectra of the methyl ester of V and of X are given. For Communication LXXXVI see KZhin., 1959, 8395. -- L. Novotny.

Card : 6/6

G-46

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds and Their Synthetic Analogues. G-3

J. Chem. Soc. : Ref Zhur-Khimiya, No 9, 1959, 31508

Androst-3 $\alpha$ -one-17 acetate (II) is produced by bromination of I in  $CCl_4$  at  $-50^\circ$ , yield 80 percent, melt. p.  $187-188^\circ$  (from alc.),  $[\alpha]_{20}^D = +55^\circ$  ( $c = 1.59$ ). The solution of II in  $CHCl_3$  is allowed to stand with methanol solution of HBr for 20 hours, and 16 $\alpha$ -bromoandrost-3 $\alpha$ -one-17 (III) is separated, yield 69 percent, melt. p.  $205-207^\circ$ ,  $[\alpha]_{20}^D = +53^\circ$  ( $c = 3.05$ ); II is obtained again by acetylation of III. 16 $\beta$ -bromoandrost-3 $\alpha$ -,17 $\beta$  triacetate is formed, with a simultaneous epimerization at C<sub>16</sub>, by the reduction of II with  $NaBH_4$  (at  $50^\circ$  in alcohol, 20 hours); the yield

Card : 2/16

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds and Their Synthetic Analogues. G-3

Abstr Jour : Ref Zhur-Khimiya, No 9, 1959, 31508

by boiling IV with methanol solution of KOH (48 hours), yield 70 percent, melt. p. 181-182° (from acetone-petr. ether),  $[\alpha]_D^{20} = +94^\circ$  (c = 2.42; alc.), which confirms the 16  $\beta$  configuration of Br. The configuration of OH at C(17) in IV and V was established by hydrogenation of V (5 percent Pd / CaCO<sub>3</sub> in alcohol) into the known androstane-3  $\alpha$ , 17  $\beta$  diacetate, melt. p. 161-162° (from CH<sub>3</sub>OH),  $[\alpha]_D^{20} = +119.4^\circ$  (c = 1.23). Oxidation of IV with CrO<sub>3</sub> in CH<sub>3</sub>COOH (20°, 20 hours) produces 16  $\beta$ -bromoandrostane-3  $\alpha$ -one-17 acetate, melt. p. 103-105° (from CH<sub>3</sub>OH),  $[\alpha]_D^{20} = +107^\circ$  (c = 1.49), which when

Card : 4/16

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds and Their Synthetic Analogues. G-3

Abs Jour : Ref Zhur-Khiniya, No 9, 1959, 31508

hours. The oily product obtained, when acetylated with  $(\text{CH}_3\text{CO})_2\text{O}$  in pyridine, produces 16 $\alpha$ -bromoandrostenediol-3 $\alpha$ , 17 $\beta$ -diacetate (II), melt. p. 171-172 $^\circ$  (from  $\text{CH}_3\text{OH}$ ),  $[\alpha]_{20\text{D}} = -29^\circ$  (c = 2.07). The 16 $\alpha$  configuration of Br in IX is confirmed by dehydrobromination (boiling with methanol KOH for 72 hours) into 16 $\beta$ , 17 $\beta$ -epoxyandrostanol-3 $\alpha$ , yield 82 percent, melt. p. 190-191 $^\circ$  (after crystallization from  $\text{CH}_3\text{OH}$  and sublimation at 180 $^\circ$  / 0.1 mm),  $[\alpha]_{20\text{D}} = +35.5^\circ$  (c = 1.69); acetate, melt. p. 185-188 $^\circ$  (from  $\text{CH}_3\text{OH}$ ),  $[\alpha]_{20\text{D}} = +289$  (c = 1.85). By the saponi-

Card : 6/16

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds and Their Synthetic Analogues. G-3

Abstr Jour : Ref Zhur-Khiniya, No 9, 1959, 31508

203° (from CH<sub>3</sub>OH),  $[\alpha]_{20D} = +113^{\circ}$  (c = 2.39) is obtained, which when oxidized with CrO<sub>3</sub> in CH<sub>3</sub>COOH, produce VIII, yield 50 percent. Analogous conversions were carried out with dehydroepiandrosterous acetate.  $\Delta^{5,16}$ -androstadienediol-3 $\beta$ -17 diacetate is obtained from the latter (similarly to I), yield 48 percent, melt. p. 147-148° (from benz.-petr. ether),  $[\alpha]_{20D} = -49.7^{\circ}$  (c = 3.82); when brominated in CCl<sub>4</sub> at -10°, this diacetate produces  $\Delta^{5,16}$   $\alpha$ -bromo-androstenol-3 $\beta$ -one-17 acetate (XII), yield 77 percent, melt. p. 181-183° (from alc.),  $[\alpha]_{20D} = -23.7^{\circ}$  (c = 3.77). By the

Card : 8/16

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds G-3  
and Their Synthetic Analogues.

Abs Jour : Ref Zhur-Khimiya, No 9, 1959, 31508

$[\alpha]_{20D} = +9^\circ$  ( $c = 2.22$ ). By the saponification of XV,  $\Delta^{5-16/\beta}$ -bromo-androstenediol-3 $\beta$ ,17 $\beta$  is obtained, melt. p. 175-176 $^\circ$  (from CH<sub>3</sub>OH),  $[\alpha]_{20D} = -41.5^\circ$  ( $c = 1.87$ ). The dehydrobromination of VI (methanol solution of KOH), 20 $^\circ$ , 48 hours) results in dehydroepiandrosterone (XVI), yield 67 percent, melt. p. 147-148 $^\circ$  (from CH<sub>3</sub>OH),  $[\alpha]_{20D} = +2.8^\circ$  ( $c = 1.75$ ), which confirms the 16 $\beta$ -configuration of Br. By the oxidation of VI with CrO<sub>3</sub> in pyridine,  $\Delta^{5-16/\beta}$ -bromoandrostenediol-3 $\beta$ -one-17-acetate (XVII) is produced, yield 69 percent, melt. p. 173-174 $^\circ$  (from CH<sub>3</sub>OH),

Card : 10/16

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds and Their Synthetic Analogues. G-3

Abs Jour : Ref Zhur-Khiniya, No 9, 1959, 31508

separated from this mixture after acetylation with  $(\text{CH}_3\text{CO})_2\text{O}$  in pyridine and crystallization from  $\text{CH}_3\text{OH}$ . XX: yield 33 percent, melt. p. 186-188,  $[\alpha]_{20\text{D}} = -94^\circ$  (c = 2.56); XXI: yield 15 percent, melt p. 203-205,  $[\alpha]_{20\text{D}} = -64^\circ$  (c = 1.34). The dehydrobromination of XXI results in XVI, yield 62 percent, and from XX  $\Delta^{5-16/3}, 17/3$ -opoxyandrost-3/3 is formed, melt. p. 143-144 (from  $\text{CH}_3\text{OH}$ ),  $[\alpha]_{20\text{D}} = -25^\circ$  (c = 1.84); acetate, melt. p. 168-169 (from  $\text{CH}_3\text{OH}$ ),  $[\alpha]_{20\text{D}} = -68.5^\circ$  (c = 1.11). XIV and XIX diketones were used to obtain epimeric 16-bromotestosterones. The reduction

Card : 12/18

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds G-3  
and Their Synthetic Analogues.

Abstr Jour : Ref Zhur-Khimiya, No 9, 1959, 31508

of XIV with  $\text{LiAlH}_4$  at  $0^\circ$  produces crude  
 $\Delta^4$ -16 $\alpha$ -bromoandrostenediol-3 $\beta$ , 17 $\beta$   
(XXII), which is oxidized in  $\text{CHCl}_3$  with  
freshly prepared  $\text{MnO}_2$  ( $20^\circ$ , 18 hours) into  
 $\Delta^4$ -16 $\alpha$ -bromoandrostenol-17 $\beta$ -one-3  
(XXIII), yield 21 percent, melt. p. 172-174 $^\circ$   
(from chloroform-petr. ether),  $[\alpha]_{20D} = +53.8^\circ$   
(c = 2.6); acetate, melt. p. 216-217 $^\circ$  (from  
 $\text{CH}_3\text{OH}$ ),  $[\alpha]_{20D} = +29.8^\circ$  (c = 1.41).  $\Delta^4$ -  
-16 $\beta$ -bromoandrostenol-17 $\beta$ -one-3 (XXV)  
is synthesized analogously to XXIII from  
XIX through the crude 16-epimer of XXII  
(XXIV), yield 26 percent, melt. p. 187-189 $^\circ$   
(from acetone),  $[\alpha]_{20D} = +73^\circ$  (c = 2.14 $^\circ$ );

Card : 13/16

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds G-3  
and Their Synthetic Analogues.

Abs Jour : Ref Zhur-Khiniya, No 9, 1959, 31508

acetate, melt. p. 158-159° (from CH<sub>3</sub>OH),  
[α]<sub>D</sub><sup>20</sup> = +126° (c = 2.41). The configura-  
tion of Br in XXIII and XXV is confirmed by  
their oxydation with CrO<sub>3</sub> in pyridine into  
XIV, yield 65 percent, and XIX, yield 67  
percent respectively. The dehydrobromina-  
tion of XXII (boiling with methanol KOH for  
48 hours) resulted in a product, the infra-  
red spectrum of which did not show the CO  
group; on the contrary, the product of de-  
hydrobromination of XXIV contained the CO  
group, which confirms the configuration of  
OH in the initial compounds. XXV was synthe-  
sized also in another way. XVIII formate,

Card : 14/16

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds G-3  
and Their Synthetic Analogues.

Abs Jour : Ref Zhur-Khiniya, No 9, 1959, 31508

melt. p. 190-191° (from CH<sub>3</sub>OH),  $[\alpha]_{20}^D = +45.5$  percent (c = 1.82) was obtained by heating XVIII in 87 percent HCOOH;  $\Delta^5$ -16 $\beta$ -bromodiol-3 $\beta$ ,17 $\beta$  3-formate (XXIV) is separated from the latter with LiAlH<sub>4</sub>-[OC(CH<sub>3</sub>)<sub>3</sub>]<sub>3</sub> (at 0°, 10 min), yield 97 percent, melt. p. 174-175° (from benz.),  $[\alpha]_{20}^D = -59^\circ$  (c = 2.42). XXVI can also be synthesized with a 30 percent yield by partial formylation of XIX, heating with 87 percent HCOOH for 15 min. Acetylation of XXVI with (CH<sub>3</sub>CO)<sub>2</sub>O in pyridine results in 3-formate-17-acetate of  $\Delta^5$ -16 $\beta$ -bromoandrostenediol-3 $\beta$ ,17 $\beta$ , yield 81 percent, melt.

Card : 15/16

165

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds G-3  
and Their Synthetic Analogues.

Abs Jour : Ref Zhur-Khimiya, No 9, 1959, 31508

p. 200-201<sup>o</sup> (from chlrof.-CH<sub>3</sub>OH),  $[\alpha]^{20}_D = +5.2^o$  (c = 2.58). The latter produces XXV acetate by the oxidation with Al isopropylate in toluene in the presence of cyclohexanone, and the XXV acetate produces XXV by saponification. Data concerning the ultraviolet spectra of XIV, XIX, XXIII and XXV are presented. All the  $[\alpha]_D$ s were determined in CHCl<sub>3</sub>, the exceptions are mentioned. See Ref Zhur-Khimiya, 1958, 81783, for report XXXIV. -- Antonin Eir.

Card : 16/16

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khiniya, No 7, 1959, 2:569

Author : Farkas, J., Kourin, P., ~~Sorn, E.~~

Inst : -

Title : The Relationship between Chemical Structure  
and Insecticidal Activity of Pyrethroid Com-  
pounds. II. Analogues of Chrysanthemic Acid  
Containing Atoms of Chlorine in the Side  
Chain.

Orig Pub : Chem. listy, 1958, 52, No 4, 688-694

Abstract : For the purpose of investigating the rela-  
tionship between the insecticidal activity  
and structure, cis-(Ia) and trans-2-( $\beta\beta$ ) -

Card : 1/9

H-90

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24569

dichlorovinyl)-3,3-dimethylcyclopropane  
carbonic acids (Ib), and also ester of  
Ib and 2-allyl-3-methyl-4-oxy-2-cyclopen-  
tenone (allylretolon) (II) were synthe-  
sized. By the addition of  $\text{CCl}_4$  to 3-methyl-  
butane-1 in the presence of benzoyl pero-  
xide (23 hours in an autoclave at  $90-95^\circ$ )  
a 38 percent yield of 1,1,1,3-tetrachlor-  
-4-methylpentane (III) of  $80-82^\circ/10$  mm  
boiling point and  $1.4860 n_D^{20}$  was obtai-  
ned. In the dehydrochlorination of alco-  
hol solution of III with 1 mol of KOH  
(48 hours at  $0^\circ$ ), 1,1,1-trichlor-4-methyl-

Card : 2/9

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24569

pentene-2 (IV) of 78-83°/30 mm and of  
1.4800 n<sub>D</sub><sup>20</sup> is being formed. In the pre-  
sence of 2 mols of KOH (48 hours at 20°)  
III yields a mixture containing 90 percent  
of IV (or also 1,1,3-trichlor-4-methylpen-  
tene-1) and 10 percent of 1,1-dichlor-3-  
-ethoxy-4-methylpentene-1. The acetylation  
of 1,1,1-trichlor-2-oxy-4-methylpentene 3  
(Ref Zhur-Khimiya, 1957, 37259) while hea-  
ting with acetic anhydride in C<sub>6</sub>H<sub>6</sub>N (3  
hours up to 100°) results in the 86 per-  
cent yield of 1,1,1-trichlor-2 acetoxy-4-  
-methylpentene-3 (V) of 98°/10 mm boiling

Card : 3/9

H-91

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24569

point and 1.4795  $n_{20}^D$ . Analogically, from a mixture of 1,1,1-trichlor-2-oxy-4-methylpentene-3 and -pentene-4, obtained through condensation of chloral and iso-butylene in accordance with Callonge and Perro (Ref Zhur-Khimiya, 1955, 26089; 1957, 54285) forms a mixture of V and 1,1,1-trichlor-2-acetoxy-4-methylpentene-4 (VI) with 89 percent yield. By introducing an ether solution of V drop by drop (in the course of 3 hours) to the suspension of Zn dust in the boiling mixture of ether and  $\text{CH}_3\text{COOH}$ , 76 percent yield of 1,1-dichlor-4-methylpenta-

Card : 4/9

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Hastáides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24569

diene-1,3 (VII) of 165-170°/720 mm boiling point,  $n_{D}^{20}$  1.5199,  $lg \epsilon_{max}$  of 4.87 at 253 m $\mu$  is derived. Analogically, by conducting the reduction of a mixture of V and VI, 62 percent yield of a mixture containing VII and its unconjugated isomer (of 145-150°/720 mm boiling point of a mixture  $n_{D}^{20}$  1.4798) is obtained, which, isomerized by heating for 1 hour with n-toluene sulfonic acid up to 150°, results in 81 percent yield of VII. When mixture of 0.3 mols of VII and 0.2 mols of diazoacetic ester are added drop by drop in the

Card : 5/9

H-92

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24569

course of 4 hours to 0.5 g of Cu-dust, heated to 110°, followed by filtering-out the catalyst and subjecting the filtrate to fractional distillation, 17.52 g of a mixture containing ethyl esters of Ia and Ib acids is obtained, having 119-120°/15 mm boiling point and 1.4883 n<sup>20</sup><sub>D</sub>. Purest sample of the mixture [with 130-135°/15 mm boiling point (bath temperature) and 1.4907 n<sup>20</sup><sub>D</sub>] is obtained after hydrolysis and esterification of the Ia and Ib mixtures (La Forge, F. B., Berthl, W. F., J. org. Chem., 1947, 12, 199). The purified mixture of esters (15.16 g) is

Card : 6/9

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24569

heated for 4 hours at 100° in 80 ml CH<sub>3</sub>COOH  
and 15 ml of 20 percent HCl (acid), followed  
by pouring into water, extraction with petro-  
leum ether, fractionation, yielding 10.5 g  
of viscous oil that has 100-110° /0.2 boiling  
point, which after mixing with 10 ml of n-  
hexane produces 7.46 g of Ia and Ib mixture  
of 60-65° melting point. By agitating 2.8 g  
of the preceding mixture in 20 ml of he-  
xane, 0.85 g of Ib having 95-96.5° melting  
point (from hexane) are obtained. From mo-  
ther liquors obtained after the separation  
of Ib, upon extended standing at -30°, 0.15 g

Card : 7/9

H-93

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24569

of Ia having 88-89° melting point (from Hexane) are derived. By heating 0.277 g Ib with 1 ml SOCl<sub>2</sub> in 3 ml of n-hexane (1 hour on a steam bath), followed by dissolving of raw chloranhydride in 5 ml C<sub>6</sub>H<sub>6</sub>, addition of 0.2 g II, and 12 hours standing of the mixture, 0.2 g of a complex ester of Ib acid with II are obtained with the boiling point of 140-150/0.2 mm and n<sub>D</sub><sup>20</sup> of 1.5274. In conducting comparison tests of insecticidal activities of Ib ester and II as against that of al-letrine (on ordinary house flies), it has been established that substitution of methyl

Card : 8/9

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24569

groups in the side chain of the chrysanthemic acid with chlorine atoms does not lead to changes in the insecticidal activity of a compound.

Card : 9/9

H-94

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24570

Author : Farkas, J., Kourim, P., ~~Sorm, F.~~

Inst : -

Title : The Relationship between Chemical Structure  
and Industrial Activity of the Pyrethroid  
Compounds. II. Analogues of Chrysanthemic  
Acid Containing Aryl Group.

Orig Pub : Chem. listy, 1958, 52, No 4, 695-706

Abstract : By the condensation of diazoacetic ester  
with the substituted derivation of styrol,  
2-arylcyclopropane carbonic acids and the  
esters with alletrolon (I) are obtained.

Card : 1/12

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24570

In the comparison of insecticidal activities of these esters with those of alle-trene, it was established that the substitution of iso-butylene in the chrysanthemic acid (II) for a phenyl group does not lead to the lowering of its activity. However, its activity is lowered when the phenyl group is being substituted. The presence of hemin  $\text{CH}_2$ -groups in the cyclopropane ring of II is essential from the standpoint of insecticidal activity. The most effective of all the esters obtained are ( $\pm$ )-trans-2-phenyl-33-dimethylcyclopropanecarbonic acid

Card : 2/12

H-95

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24570

ester (trans-III) or its pravoratory anti-  
podes (trans- (+) -III) with I. The abso-  
lute configuration of trans-III has been  
proved. The following aromatic carbinols  
 $R_1R_2R_3 \cdot COH$  are obtained (using usual  
methods) from an aldehyde, ketone, or com-  
pound ester and aryl, or alkyl-MgBr, or  
the corresponding MgCl derivatives. Presen-  
ted are:  $R_1, R_2, R_3$ , boiling point in  $^{\circ}C/mm$ ,  
 $n_D^{20}$ , yield in percent. They are: H,  $C_2H_5$ ,  
 $C_6H_6$  (IV), 110/15, 1.5262, 66;  $CH_3$ ,  $CH_3$ ,  
benzyl, 95/8, 1.5169, 74;  $C_2H_5$ ,  $C_2H_5$ , ben-  
zyl, 125/15, 1.5165, 64;  $CH_3$ ,  $CH_3$ ,  $\delta$ -xylyl;

Card : 3/12

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24570

113/8, 1.5186, 71;  $\text{CH}_3\text{CH}_2$ , m-xylyl, 119/20,  
1.5144, 71;  $\text{CH}_3$ ,  $\text{CH}_2$ , n-xylyl IV, 102/8  
(melting point 90-418 [sic]), 1.5129, 73  
3,5-dinitrobenzoate IV, melting point of  
133-134°;  $\text{CH}_3$ ,  $\text{CH}_2$ , n-chlorbenzyl (V),  
129/11 (340 melting point), -, 73 (3,5-  
-dinitrobenzoate V, melting point 126-127°);  
 $\text{CH}_3$ ,  $\text{CH}_2$ , 2,4-dichlorobenzyl, 135/10, 1.5451,  
71 (3,5-dinitrobenzoate, melting point 116°).  
The following derivatives of styrol  $\text{R}_1 \cdot \text{C}_6\text{H}_4 \cdot$   
•  $\text{CH} = \text{C}(\text{R}_2)(\text{R}_3)$  were obtained in the 3 hours  
of boiling of the corresponding alcohol with  
30 percent excess of acetic anhydride and 1

Card : 4/12

H-96

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 29570

percent  $H_2SO_4$  (Method A), or from the  
corresponding alcohol in the contact with  
 $Al_2O_3$  at  $300-320^\circ$  (Method B). Presented  
below are R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, boiling point in  $^\circ C/$   
 $/mm$ ,  $n_{20D}$ , synthesis method, yield in per-  
cent. They are: H, H,  $CH_3$  (VI), 71/15,  
1.5485, A, 52; H,  $CH_3$ ,  $CH_3$ , 78/16, 1.5387,  
A, 82; H,  $C_2H_5$ ,  $C_2H_5$  (VII) 101/15, 1.5189,  
A, 86; o- $CH_3$ ,  $CH_3$ ,  $CH_3$ , 87/18, 1.5283, A,  
84; m- $CH_3$ ,  $CH_3$ ,  $CH_3$ , 92/17, 1.5312, A, 84;  
n- $CH_3$ ,  $CH_3$ ,  $CH_3$ , 97/22, 1.5332, A, 88 (B,  
66); n-Cl,  $CH_3$ ,  $CH_3$ , 102/11, 1.5521, A, 74;  
2,4-Cl<sub>2</sub>,  $CH_3$ ,  $CH_3$ , 123/15, 1.5593, A, 89.

Card : 5/12

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24570

VI ( $n_{20D} = 1.5490$ ) also obtained with 80 percent yield from allylbenzene by heating it with 1 percent tert-butylate K up to  $150^{\circ}$ , but VII ( $n_{20D} = 1.5314$ ) with 55 percent yield by heating (1 hour) 1-methyl-3-methylbenzene with 1 percent n-toluenesulfonic acid at  $160^{\circ}$  temperature. 2-arylcyclopropanecarboxylic acids  $R_1C_6H_4CHC(R_2)(R_3)CHCOOH$  were synthesized when a mixture of 0.2-0.3 mols of corresponding styrol derivative and 0.1 mols of diazoacetic ester were added drop by drop to 0.1 mols of styrol derivative and 1 g Cu dust in the course of 6 hours at  $120-125^{\circ}$

Card : 6/12

H-97

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24570

with the subsequent heating for  $1/2$  hour at  $150^{\circ}$ . The mixture of corresponding ethyl esters, isolated by vacuum distillation, is saponified by boiling for 3 hours with 20 percent water-alcohol solution of KOH (50 percent excess). From the mixture of stereoisomeric acids, after the addition of petroleum ether, the corresponding cis-acid in crystalline form (notations of cis- and trans- refer to corresponding positions of  $R_1 \cdot C_6H_4$  and COOH groups) is usually separated. The non-crystallizing mixtures are converted (by means of 3 hour heating with  $SOCl_2$ ) into  $C_6H_6$

Card : 7/12

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24570

with the corresponding chloranhydrides, or  
else by the interaction of these chloran-  
hydrides with  $NH_3$  solution in  $CHCl_3$  at  $0^\circ$ ,  
into the corresponding amides. All the cis-  
acids were recrystallized from  $C_6H_6$  + petro-  
leum ether, all the trans-acids from petro-  
leum ether. The following arylcyclopropane acids  
were obtained (given below are  $R_1, R_2, R_3$ ,  
yield in percent of the isomeric acids, mel-  
ting point in  $^\circ C$  of trans- and cis-isomers,  
boiling point in  $^\circ C/mm$  of chloranhydrides of  
cis- and trans- acids): H, H, H VIII, 59, 104,  
92, -, -, (melting points of amides cis-VIII

Card : 8/12

H-98

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24570

and trans-VIII are 88-89° and 189°, melting  
point of toluidido cis-VIII is 174°); H, H,  
CH<sub>3</sub> (IX), 50, 114, 77.5, -, -; H, CH<sub>3</sub>, CH<sub>3</sub>  
(III), 63, 134, 102, -, 122/11 (melting  
points of amides cis- and trans-III are 98-  
99° and 141°); 2-CH<sub>3</sub>, CH<sub>3</sub>, CH<sub>3</sub> (X), 40, -,  
126, -, 144/16; 3-CH<sub>3</sub>, CH<sub>3</sub> (XI), 62, -, boi-  
ling point 135°/0.1 mm, -, 122/5 (melting  
point of toluidido trans-XI is 145-147°);  
4-CH<sub>3</sub>, CH<sub>3</sub>, CH<sub>3</sub> (XII), 62, 142, 134, -,  
105/0.4; 4-Cl, CH<sub>3</sub>, CH<sub>3</sub> (XIII), 46, -,  
140-141; -, 120/0.2; 2,4-Cl<sub>2</sub>, CH<sub>3</sub>, CH<sub>3</sub>, 46,  
174 (?), 96 (?), -, -; H, C<sub>2</sub>H<sub>5</sub>, C<sub>2</sub>H<sub>5</sub> (XIV),

Card : 9/12

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24570

58, -, 98, -, -. When a mixture of solutions of 9.5 g of trans-III in 50 ml of ether and 4.2 g (-) ephedrin in 10 ml of ether are kept at 0° for 3 hours, 5.85 g (-)-ephedrin salt of trans-(+)-III of 169° melting point (from ethylacetate),  $[\alpha]_{20}^D$  of +0.51° (with 3.13, in C<sub>2</sub>H<sub>5</sub>OH) are derived. From mother liquors after addition of 4.9 g (-)-ephedrin and after keeping at 0° for 5 hours, (-)-ephedrin salt of trans-(+)-III of 130-131° melting point (from ethylacetate),  $[\alpha]_{20}^D = 38.8°$  (with 2.64 in C<sub>2</sub>H<sub>5</sub>OH) is obtained. And from the decom-

Card : 10/12

H-99

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticides.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24570

position of salts with 4 percent HCl  
(acid), were obtained free trans-(+)-III  
of 84-85° melting point,  $[\alpha]_{20D} + 31.9^\circ$   
(with 4.93 in  $C_2H_5OH$ ) and trans-(-)-III of  
84-85.5° melting point,  $[\alpha]_{20D} - 32.2^\circ$   
(with 4.85, in  $C_2H_5OH$ ). Esters of the pre-  
ceding acids and I are obtained with yields  
of approximately 50 percent in a mixture of  
 $C_6H_6 + C_2H_5N$  from I and chloranhydride of  
corresponding acid when keeping a mixture  
for 1 hour at 0° and 12 hours at approxima-  
tely 20°. After the usual treatment of a  
solution in  $C_6H_6$ , the raw ester is purified

Card : 11/12

CZECHOSLOVAKIA/Chemical Technology. Chemical  
Products and Their Applications.  
Pesticidos.

H-18

Abs Jour : Ref Zhur-Khimiya, No 7, 1959, 24570

with  $Al_2O_3$  and volatile fractions are re-  
moved by heating to  $78^{\circ}/0/1$  mm for 8 hours,  
or by fractionation at  $180-190^{\circ}/0/1$  mm. The  
following esters were synthesized (given be-  
low are initial acid and  $n_{20}^D$  of ester): trans-  
-VIII, 1.5517; trans-DX, 1.5450; trans-III,  
1.5382; trans-(+)-III, 1.5371; trans-(-)-III,  
1.5378; cis-III, 1.5401; X, 1.5369; XI, 1.5362;  
trans-XII, 1.5365; XIII, 1.5452; XIV, 1.5355.  
For Part I, see preceding abstract.

Card : 12/12

H-100

SORM, F.

CHEMISTRY/Organic Chemistry. Natural Compounds and Their Synthetic Analogs.

The Jour: Inf Bur-Div., No 11, 1979, 3675.

Author: I. G. Kopylov, I. I. Kopylov, D. I. Kopylov, V. I. Kopylov, M. I. Kopylov, S. I. Kopylov, P. I. Kopylov.

Title: Chemistry of the Terpenes. I. Structure of Ginkgolides, the Cerebrolides Component of Ginkgolide Medicinal Preparation.

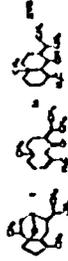
ORAC. No: Chem Libr., 22, No 6, 1163-1173 (1973) (in Czech)

Abstract: The authors have shown that the previously proposed structure of ginkgolide medicinal essential oil (Ginkgolide) is incorrect. They have the cyclic structure (I) (see insert below), as

Card: 1/12

CHEMISTRY/Organic Chemistry. Natural Compounds and Their Synthetic Analogs.

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CHEMISTRY/Organic Chemistry. Natural Compounds and Their Synthetic Analogs.

The Jour: Inf Bur-Div., No 11, 1979, 3675.

previously assumed, but that of the ketone (II). The authors propose the new structure for II. The structure of II has been proved experimentally and by chemical reactions. The reduction of I in classical conditions (LiAlH<sub>4</sub>) with LiCl gives a liquid product (II) which is identical with the liquid product (II) obtained with LiAlH<sub>4</sub> in the presence of LiCl. The oxidation of II with CrO<sub>3</sub> in LiCl does not form a ketone (III) but a hydroxyketone (IV). The hydroxyketone (IV) is identical with the hydroxyketone (IV) obtained by the oxidation of I by alkaline isopropylate or by LiAlH<sub>4</sub> in the presence of LiCl. The hydroxyketone (IV) yields a hydration (V). The dehydration of V

Card: 1/12

COMMUNION/II/Organic Chemistry, Natural Compounds and Their Synthetic Analogs.

The Jour: Int Zhur-Khim., No 11, 1959, 3877.

or of VI with 8 gives octalones (VII). The hydro-  
genation of VI in classical CH<sub>3</sub>COOH over Pt (from PtO<sub>2</sub>)  
with a solution of H<sub>2</sub> gives alcohols (VIII); the hydro-  
genation of V under the same conditions gives solutions  
(IX). The hydrogenation of I over Pt (from PtO<sub>2</sub>) in  
alcohol gives 6, 7, 8-trimethyl-2,3-dihydrocyclo-  
hexane (X). The reduction of I by LiAlH<sub>4</sub> gives tetra-  
hydrofuran (XI) which on hydrogenation in classical  
CH<sub>3</sub>COOH over Pt (from PtO<sub>2</sub>) gives the saturated cyclic  
carbon paraffins (XII) and XIII. The IR spectrum  
of XII differs from the spectra of saturated aliphatic  
paraffin hydrocarbons and resembles that of cyclohexane  
and furanones (see Izv. Akad. Nauk SSSR, 1953, 5644). The substitu-

Cont : 4/12

0-49

tion of III gives a hydrocarbon (XIII). The oxida-  
tion of I yields 1 mol of acetone, while the oxidation of  
I gives 1,5-1,6 mol of acetone, the oxidation of  
I by MnO<sub>4</sub> in acetic acid (COOH) and levulinic  
acid. The action of O<sub>2</sub> gives on I, in the cold gives  
crystalline tetramers (XIV). The hydrogenation  
of XIV in classical CH<sub>3</sub>COOH over Pt (from PtO<sub>2</sub>) gives  
IV; when the hydrogenation is carried out in alcohol,  
tetrahydrofuran (XI) is obtained. The acid  
isomerization of I yields a liquid ketone (XVI) which  
on hydrogenation over Pt (from PtO<sub>2</sub>) in classical CH<sub>3</sub>  
COOH with subsequent reduction by LiAlH<sub>4</sub> gives alcoh-  
ols (XVII) (cf. manuscript); oxidation of the

Cont : 5/12

latter product yields octalones (XVIII) [5].  
The degradation of XVII followed by hydrogenation  
of the product yields 1,5-1,6 mol of acetone over  
(from PtO<sub>2</sub>) [5] (278 cc) in classical CH<sub>3</sub>COOH over  
hydrogenation over PtO<sub>2</sub> gives II, 2, 3, 4, 5, 6, 7,  
8 mol of acetone, 0.9152. The reduction of  
1,4,5,6,7,8,9,10,11,12,13,14,15,16,17,18,19,20,21,22,23,24,25,26,27,28,29,30,31,32,33,34,35,36,37,38,39,40,41,42,43,44,45,46,47,48,49,50,51,52,53,54,55,56,57,58,59,60,61,62,63,64,65,66,67,68,69,70,71,72,73,74,75,76,77,78,79,80,81,82,83,84,85,86,87,88,89,90,91,92,93,94,95,96,97,98,99,100,101,102,103,104,105,106,107,108,109,110,111,112,113,114,115,116,117,118,119,120,121,122,123,124,125,126,127,128,129,130,131,132,133,134,135,136,137,138,139,140,141,142,143,144,145,146,147,148,149,150,151,152,153,154,155,156,157,158,159,160,161,162,163,164,165,166,167,168,169,170,171,172,173,174,175,176,177,178,179,180,181,182,183,184,185,186,187,188,189,190,191,192,193,194,195,196,197,198,199,200,201,202,203,204,205,206,207,208,209,210,211,212,213,214,215,216,217,218,219,220,221,222,223,224,225,226,227,228,229,230,231,232,233,234,235,236,237,238,239,240,241,242,243,244,245,246,247,248,249,250,251,252,253,254,255,256,257,258,259,260,261,262,263,264,265,266,267,268,269,270,271,272,273,274,275,276,277,278,279,280,281,282,283,284,285,286,287,288,289,290,291,292,293,294,295,296,297,298,299,300,301,302,303,304,305,306,307,308,309,310,311,312,313,314,315,316,317,318,319,320,321,322,323,324,325,326,327,328,329,330,331,332,333,334,335,336,337,338,339,340,341,342,343,344,345,346,347,348,349,350,351,352,353,354,355,356,357,358,359,360,361,362,363,364,365,366,367,368,369,370,371,372,373,374,375,376,377,378,379,380,381,382,383,384,385,386,387,388,389,390,391,392,393,394,395,396,397,398,399,400,401,402,403,404,405,406,407,408,409,410,411,412,413,414,415,416,417,418,419,420,421,422,423,424,425,426,427,428,429,430,431,432,433,434,435,436,437,438,439,440,441,442,443,444,445,446,447,448,449,450,451,452,453,454,455,456,457,458,459,460,461,462,463,464,465,466,467,468,469,470,471,472,473,474,475,476,477,478,479,480,481,482,483,484,485,486,487,488,489,490,491,492,493,494,495,496,497,498,499,500,501,502,503,504,505,506,507,508,509,510,511,512,513,514,515,516,517,518,519,520,521,522,523,524,525,526,527,528,529,530,531,532,533,534,535,536,537,538,539,540,541,542,543,544,545,546,547,548,549,550,551,552,553,554,555,556,557,558,559,560,561,562,563,564,565,566,567,568,569,570,571,572,573,574,575,576,577,578,579,580,581,582,583,584,585,586,587,588,589,590,591,592,593,594,595,596,597,598,599,600,601,602,603,604,605,606,607,608,609,610,611,612,613,614,615,616,617,618,619,620,621,622,623,624,625,626,627,628,629,630,631,632,633,634,635,636,637,638,639,640,641,642,643,644,645,646,647,648,649,650,651,652,653,654,655,656,657,658,659,660,661,662,663,664,665,666,667,668,669,670,671,672,673,674,675,676,677,678,679,680,681,682,683,684,685,686,687,688,689,690,691,692,693,694,695,696,697,698,699,700,701,702,703,704,705,706,707,708,709,710,711,712,713,714,715,716,717,718,719,720,721,722,723,724,725,726,727,728,729,730,731,732,733,734,735,736,737,738,739,740,741,742,743,744,745,746,747,748,749,750,751,752,753,754,755,756,757,758,759,760,761,762,763,764,765,766,767,768,769,770,771,772,773,774,775,776,777,778,779,780,781,782,783,784,785,786,787,788,789,790,791,792,793,794,795,796,797,798,799,800,801,802,803,804,805,806,807,808,809,810,811,812,813,814,815,816,817,818,819,820,821,822,823,824,825,826,827,828,829,830,831,832,833,834,835,836,837,838,839,840,841,842,843,844,845,846,847,848,849,850,851,852,853,854,855,856,857,858,859,860,861,862,863,864,865,866,867,868,869,870,871,872,873,874,875,876,877,878,879,880,881,882,883,884,885,886,887,888,889,890,891,892,893,894,895,896,897,898,899,900,901,902,903,904,905,906,907,908,909,910,911,912,913,914,915,916,917,918,919,920,921,922,923,924,925,926,927,928,929,930,931,932,933,934,935,936,937,938,939,940,941,942,943,944,945,946,947,948,949,950,951,952,953,954,955,956,957,958,959,960,961,962,963,964,965,966,967,968,969,970,971,972,973,974,975,976,977,978,979,980,981,982,983,984,985,986,987,988,989,990,991,992,993,994,995,996,997,998,999,1000,1001,1002,1003,1004,1005,1006,1007,1008,1009,1010,1011,1012,1013,1014,1015,1016,1017,1018,1019,1020,1021,1022,1023,1024,1025,1026,1027,1028,1029,1030,1031,1032,1033,1034,1035,1036,1037,1038,1039,1040,1041,1042,1043,1044,1045,1046,1047,1048,1049,1050,1051,1052,1053,1054,1055,1056,1057,1058,1059,1060,1061,1062,1063,1064,1065,1066,1067,1068,1069,1070,1071,1072,1073,1074,1075,1076,1077,1078,1079,1080,1081,1082,1083,1084,1085,1086,1087,1088,1089,1090,1091,1092,1093,1094,1095,1096,1097,1098,1099,1100,1101,1102,1103,1104,1105,1106,1107,1108,1109,1110,1111,1112,1113,1114,1115,1116,1117,1118,1119,1120,1121,1122,1123,1124,1125,1126,1127,1128,1129,1130,1131,1132,1133,1134,1135,1136,1137,1138,1139,1140,1141,1142,1143,1144,1145,1146,1147,1148,1149,1150,1151,1152,1153,1154,1155,1156,1157,1158,1159,1160,1161,1162,1163,1164,1165,1166,1167,1168,1169,1170,1171,1172,1173,1174,1175,1176,1177,1178,1179,1180,1181,1182,1183,1184,1185,1186,1187,1188,1189,1190,1191,1192,1193,1194,1195,1196,1197,1198,1199,1200,1201,1202,1203,1204,1205,1206,1207,1208,1209,1210,1211,1212,1213,1214,1215,1216,1217,1218,1219,1220,1221,1222,1223,1224,1225,1226,1227,1228,1229,1230,1231,1232,1233,1234,1235,1236,1237,1238,1239,1240,1241,1242,1243,1244,1245,1246,1247,1248,1249,1250,1251,1252,1253,1254,1255,1256,1257,1258,1259,1260,1261,1262,1263,1264,1265,1266,1267,1268,1269,1270,1271,1272,1273,1274,1275,1276,1277,1278,1279,1280,1281,1282,1283,1284,1285,1286,1287,1288,1289,1290,1291,1292,1293,1294,1295,1296,1297,1298,1299,1300,1301,1302,1303,1304,1305,1306,1307,1308,1309,1310,1311,1312,1313,1314,1315,1316,1317,1318,1319,1320,1321,1322,1323,1324,1325,1326,1327,1328,1329,1330,1331,1332,1333,1334,1335,1336,1337,1338,1339,1340,1341,1342,1343,1344,1345,1346,1347,1348,1349,1350,1351,1352,1353,1354,1355,1356,1357,1358,1359,1360,1361,1362,1363,1364,1365,1366,1367,1368,1369,1370,1371,1372,1373,1374,1375,1376,1377,1378,1379,1380,1381,1382,1383,1384,1385,1386,1387,1388,1389,1390,1391,1392,1393,1394,1395,1396,1397,1398,1399,1400,1401,1402,1403,1404,1405,1406,1407,1408,1409,1410,1411,1412,1413,1414,1415,1416,1417,1418,1419,1420,1421,1422,1423,1424,1425,1426,1427,1428,1429,1430,1431,1432,1433,1434,1435,1436,1437,1438,1439,1440,1441,1442,1443,1444,1445,1446,1447,1448,1449,1450,1451,1452,1453,1454,1455,1456,1457,1458,1459,1460,1461,1462,1463,1464,1465,1466,1467,1468,1469,1470,1471,1472,1473,1474,1475,1476,1477,1478,1479,1480,1481,1482,1483,1484,1485,1486,1487,1488,1489,1490,1491,1492,1493,1494,1495,1496,1497,1498,1499,1500,1501,1502,1503,1504,1505,1506,1507,1508,1509,1510,1511,1512,1513,1514,1515,1516,1517,1518,1519,1520,1521,1522,1523,1524,1525,1526,1527,1528,1529,1530,1531,1532,1533,1534,1535,1536,1537,1538,1539,1540,1541,1542,1543,1544,1545,1546,1547,1548,1549,1550,1551,1552,1553,1554,1555,1556,1557,1558,1559,1560,1561,1562,1563,1564,1565,1566,1567,1568,1569,1570,1571,1572,1573,1574,1575,1576,1577,1578,1579,1580,1581,1582,1583,1584,1585,1586,1587,1588,1589,1590,1591,1592,1593,1594,1595,1596,1597,1598,1599,1600,1601,1602,1603,1604,1605,1606,1607,1608,1609,1610,1611,1612,1613,1614,1615,1616,1617,1618,1619,1620,1621,1622,1623,1624,1625,1626,1627,1628,1629,1630,1631,1632,1633,1634,1635,1636,1637,1638,1639,1640,1641,1642,1643,1644,1645,1646,1647,1648,1649,1650,1651,1652,1653,1654,1655,1656,1657,1658,1659,1660,1661,1662,1663,1664,1665,1666,1667,1668,1669,1670,1671,1672,1673,1674,1675,1676,1677,1678,1679,1680,1681,1682,1683,1684,1685,1686,1687,1688,1689,1690,1691,1692,1693,1694,1695,1696,1697,1698,1699,1700,1701,1702,1703,1704,1705,1706,1707,1708,1709,1710,1711,1712,1713,1714,1715,1716,1717,1718,1719,1720,1721,1722,1723,1724,1725,1726,1727,1728,1729,1730,1731,1732,1733,1734,1735,1736,1737,1738,1739,1740,1741,1742,1743,1744,1745,1746,1747,1748,1749,1750,1751,1752,1753,1754,1755,1756,1757,1758,1759,1760,1761,1762,1763,1764,1765,1766,1767,1768,1769,1770,1771,1772,1773,1774,1775,1776,1777,1778,1779,1780,1781,1782,1783,1784,1785,1786,1787,1788,1789,1790,1791,1792,1793,1794,1795,1796,1797,1798,1799,1800,1801,1802,1803,1804,1805,1806,1807,1808,1809,1810,1811,1812,1813,1814,1815,1816,1817,1818,1819,1820,1821,1822,1823,1824,1825,1826,1827,1828,1829,1830,1831,1832,1833,1834,1835,1836,1837,1838,1839,1840,1841,1842,1843,1844,1845,1846,1847,1848,1849,1850,1851,1852,1853,1854,1855,1856,1857,1858,1859,1860,1861,1862,1863,1864,1865,1866,1867,1868,1869,1870,1871,1872,1873,1874,1875,1876,1877,1878,1879,1880,1881,1882,1883,1884,1885,1886,1887,1888,1889,1890,1891,1892,1893,1894,1895,1896,1897,1898,1899,1900,1901,1902,1903,1904,1905,1906,1907,1908,1909,1910,1911,1912,1913,1914,1915,1916,1917,1918,1919,1920,1921,1922,1923,1924,1925,1926,1927,1928,1929,1930,1931,1932,1933,1934,1935,1936,1937,1938,1939,1940,1941,1942,1943,1944,1945,1946,1947,1948,1949,1950,1951,1952,1953,1954,1955,1956,1957,1958,1959,1960,1961,1962,1963,1964,1965,1966,1967,1968,1969,1970,1971,1972,1973,1974,1975,1976,1977,1978,1979,1980,1981,1982,1983,1984,1985,1986,1987,1988,1989,1990,1991,1992,1993,1994,1995,1996,1997,1998,1999,2000,2001,2002,2003,2004,2005,2006,2007,2008,2009,2010,2011,2012,2013,2014,2015,2016,2017,2018,2019,2020,2021,2022,2023,2024,2025,2026,2027,2028,2029,2030,2031,2032,2033,2034,2035,2036,2037,2038,2039,2040,2041,2042,2043,2044,2045,2046,2047,2048,2049,2050,2051,2052,2053,2054,2055,2056,2057,2058,2059,2060,2061,2062,2063,2064,2065,2066,2067,2068,2069,2070,2071,2072,2073,2074,2075,2076,2077,2078,2079,2080,2081,2082,2083,2084,2085,2086,2087,2088,2089,2090,2091,2092,2093,2094,2095,2096,2097,2098,2099,2100,2101,2102,2103,2104,2105,2106,2107,2108,2109,2110,2111,2112,2113,2114,2115,2116,2117,2118,2119,2120,2121,2122,2123,2124,2125,2126,2127,2128,2129,2130,2131,2132,2133,2134,2135,2136,2137,2138,2139,2140,2141,2142,2143,2144,2145,2146,2147,2148,2149,2150,2151,2152,2153,2154,2155,2156,2157,2158,2159,2160,2161,2162,2163,2164,2165,2166,2167,2168,2169,2170,2171,2172,2173,2174,2175,2176,2177,2178,2179,2180,2181,2182,2183,2184,2185,2186,2187,2188,2189,2190,2191,2192,2193,2194,2195,2196,2197,2198,2199,2200,2201,2202,2203,2204,2205,2206,2207,2208,2209,2210,2211,2212,2213,2214,2215,2216,2217,2218,2219,2220,2221,2222,2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CZERNIK, V. II. / Organic Chemistry. Natural Compounds and Their Synthetic Analogs.

The Jour: Dof Znan-shim., No 11, 1979, 35775.

Reaction of 2 cm I by aluminum isopropylate (from 1.0 gm Al) gives (after chromatography) 1.35 gm V, by 120-130/0.15 in. The reduction of 0.5 gm I by lithium gives V, by 120/0.8 cm, 1.5 gm. 1.287. When 3 cm V and 7 ml of 100% alcohol are added, the product is 1.287. The reduction of 0.5 gm I by 100% alcohol gives 1.287. The reduction of 0.5 gm I by 100% alcohol gives 1.287. The reduction of 0.5 gm I by 100% alcohol gives 1.287.

Card : 7/12

The hydrogenation of 112 mg VI in 1 ml alcohol (active towards I-II) and elution with petroleum ether, gives VII, 1.287. The reduction of 0.5 gm I by 100% alcohol gives 1.287. The reduction of 0.5 gm I by 100% alcohol gives 1.287. The reduction of 0.5 gm I by 100% alcohol gives 1.287.

Card : 8/12

The hydrogenation of 112 mg VI in 1 ml alcohol (active towards I-II) and elution with petroleum ether, gives VII, 1.287. The reduction of 0.5 gm I by 100% alcohol gives 1.287. The reduction of 0.5 gm I by 100% alcohol gives 1.287. The reduction of 0.5 gm I by 100% alcohol gives 1.287.

Card : 9/12

CONSTITUTIONAL/Organic Chemistry. Natural Compounds and Their Synthetic Analogs.

The Journ. Ref. Zhar-Khin, No 11, 1979, 3877S.

to 120-125°C. The hydrogenation of XIV (700 mg) in ethyl alcohol by the same procedure as that used for I, followed by oxidation of the reaction product by CrO<sub>3</sub> in ethyl alcohol and purification by chromatography on 1103 (active towards I-IV) gives XV, mp 120-122°C, n<sub>D</sub><sup>20</sup> 1.4760, d<sub>4</sub><sup>20</sup> 0.9228. The hydrogenation of XIV (500 mg) over Pt (from PtO<sub>2</sub>) in alcohol (of procedure used for I), followed by reduction of the reaction product (active towards I-III) gives 30 mg of III, mp 120-122°C, n<sub>D</sub><sup>20</sup> 1.4728, d<sub>4</sub><sup>20</sup> 0.8970. When 2 mg of XIV are reduced by LiAlH<sub>4</sub> (0.4 g) in ether solution, IV

Card : 10/12 0-5

is obtained, mp 120-121°C, n<sub>D</sub><sup>20</sup> 1.4660 (from alcohol); mass spectrum m/e 149 (M<sup>+</sup>), 147 (M<sup>+</sup> - 2), 145 (M<sup>+</sup> - 4), 143 (M<sup>+</sup> - 6), 141 (M<sup>+</sup> - 8), 139 (M<sup>+</sup> - 10), 137 (M<sup>+</sup> - 12), 135 (M<sup>+</sup> - 14), 133 (M<sup>+</sup> - 16), 131 (M<sup>+</sup> - 18), 129 (M<sup>+</sup> - 20), 127 (M<sup>+</sup> - 22), 125 (M<sup>+</sup> - 24), 123 (M<sup>+</sup> - 26), 121 (M<sup>+</sup> - 28), 119 (M<sup>+</sup> - 30), 117 (M<sup>+</sup> - 32), 115 (M<sup>+</sup> - 34), 113 (M<sup>+</sup> - 36), 111 (M<sup>+</sup> - 38), 109 (M<sup>+</sup> - 40), 107 (M<sup>+</sup> - 42), 105 (M<sup>+</sup> - 44), 103 (M<sup>+</sup> - 46), 101 (M<sup>+</sup> - 48), 99 (M<sup>+</sup> - 50), 97 (M<sup>+</sup> - 52), 95 (M<sup>+</sup> - 54), 93 (M<sup>+</sup> - 56), 91 (M<sup>+</sup> - 58), 89 (M<sup>+</sup> - 60), 87 (M<sup>+</sup> - 62), 85 (M<sup>+</sup> - 64), 83 (M<sup>+</sup> - 66), 81 (M<sup>+</sup> - 68), 79 (M<sup>+</sup> - 70), 77 (M<sup>+</sup> - 72), 75 (M<sup>+</sup> - 74), 73 (M<sup>+</sup> - 76), 71 (M<sup>+</sup> - 78), 69 (M<sup>+</sup> - 80), 67 (M<sup>+</sup> - 82), 65 (M<sup>+</sup> - 84), 63 (M<sup>+</sup> - 86), 61 (M<sup>+</sup> - 88), 59 (M<sup>+</sup> - 90), 57 (M<sup>+</sup> - 92), 55 (M<sup>+</sup> - 94), 53 (M<sup>+</sup> - 96), 51 (M<sup>+</sup> - 98), 49 (M<sup>+</sup> - 100), 47 (M<sup>+</sup> - 102), 45 (M<sup>+</sup> - 104), 43 (M<sup>+</sup> - 106), 41 (M<sup>+</sup> - 108), 39 (M<sup>+</sup> - 110), 37 (M<sup>+</sup> - 112), 35 (M<sup>+</sup> - 114), 33 (M<sup>+</sup> - 116), 31 (M<sup>+</sup> - 118), 29 (M<sup>+</sup> - 120), 27 (M<sup>+</sup> - 122), 25 (M<sup>+</sup> - 124), 23 (M<sup>+</sup> - 126), 21 (M<sup>+</sup> - 128), 19 (M<sup>+</sup> - 130), 17 (M<sup>+</sup> - 132), 15 (M<sup>+</sup> - 134), 13 (M<sup>+</sup> - 136), 11 (M<sup>+</sup> - 138), 9 (M<sup>+</sup> - 140), 7 (M<sup>+</sup> - 142), 5 (M<sup>+</sup> - 144), 3 (M<sup>+</sup> - 146), 1 (M<sup>+</sup> - 148). The IR spectrum of IV shows absorption bands at 3000 (ν<sub>C-H</sub>), 1650 (ν<sub>C=O</sub>), 1600 (ν<sub>C=C</sub>), 1500 (ν<sub>C=C</sub>), 1450 (ν<sub>C=C</sub>), 1400 (ν<sub>C=C</sub>), 1350 (ν<sub>C=C</sub>), 1300 (ν<sub>C=C</sub>), 1250 (ν<sub>C=C</sub>), 1200 (ν<sub>C=C</sub>), 1150 (ν<sub>C=C</sub>), 1100 (ν<sub>C=C</sub>), 1050 (ν<sub>C=C</sub>), 1000 (ν<sub>C=C</sub>), 950 (ν<sub>C=C</sub>), 900 (ν<sub>C=C</sub>), 850 (ν<sub>C=C</sub>), 800 (ν<sub>C=C</sub>), 750 (ν<sub>C=C</sub>), 700 (ν<sub>C=C</sub>), 650 (ν<sub>C=C</sub>), 600 (ν<sub>C=C</sub>), 550 (ν<sub>C=C</sub>), 500 (ν<sub>C=C</sub>), 450 (ν<sub>C=C</sub>), 400 (ν<sub>C=C</sub>), 350 (ν<sub>C=C</sub>), 300 (ν<sub>C=C</sub>), 250 (ν<sub>C=C</sub>), 200 (ν<sub>C=C</sub>), 150 (ν<sub>C=C</sub>), 100 (ν<sub>C=C</sub>), 50 (ν<sub>C=C</sub>), 0 (ν<sub>C=C</sub>). The UV spectrum of IV shows absorption bands at 280 (ε 10000), 260 (ε 10000), 240 (ε 10000), 220 (ε 10000), 200 (ε 10000), 180 (ε 10000), 160 (ε 10000), 140 (ε 10000), 120 (ε 10000), 100 (ε 10000), 80 (ε 10000), 60 (ε 10000), 40 (ε 10000), 20 (ε 10000), 0 (ε 10000). The NMR spectrum of IV shows absorption bands at 7.5 (s, 1H), 7.0 (s, 1H), 6.5 (s, 1H), 6.0 (s, 1H), 5.5 (s, 1H), 5.0 (s, 1H), 4.5 (s, 1H), 4.0 (s, 1H), 3.5 (s, 1H), 3.0 (s, 1H), 2.5 (s, 1H), 2.0 (s, 1H), 1.5 (s, 1H), 1.0 (s, 1H), 0.5 (s, 1H), 0 (s, 1H). The mass spectrum of IV shows absorption bands at 149 (M<sup>+</sup>), 147 (M<sup>+</sup> - 2), 145 (M<sup>+</sup> - 4), 143 (M<sup>+</sup> - 6), 141 (M<sup>+</sup> - 8), 139 (M<sup>+</sup> - 10), 137 (M<sup>+</sup> - 12), 135 (M<sup>+</sup> - 14), 133 (M<sup>+</sup> - 16), 131 (M<sup>+</sup> - 18), 129 (M<sup>+</sup> - 20), 127 (M<sup>+</sup> - 22), 125 (M<sup>+</sup> - 24), 123 (M<sup>+</sup> - 26), 121 (M<sup>+</sup> - 28), 119 (M<sup>+</sup> - 30), 117 (M<sup>+</sup> - 32), 115 (M<sup>+</sup> - 34), 113 (M<sup>+</sup> - 36), 111 (M<sup>+</sup> - 38), 109 (M<sup>+</sup> - 40), 107 (M<sup>+</sup> - 42), 105 (M<sup>+</sup> - 44), 103 (M<sup>+</sup> - 46), 101 (M<sup>+</sup> - 48), 99 (M<sup>+</sup> - 50), 97 (M<sup>+</sup> - 52), 95 (M<sup>+</sup> - 54), 93 (M<sup>+</sup> - 56), 91 (M<sup>+</sup> - 58), 89 (M<sup>+</sup> - 60), 87 (M<sup>+</sup> - 62), 85 (M<sup>+</sup> - 64), 83 (M<sup>+</sup> - 66), 81 (M<sup>+</sup> - 68), 79 (M<sup>+</sup> - 70), 77 (M<sup>+</sup> - 72), 75 (M<sup>+</sup> - 74), 73 (M<sup>+</sup> - 76), 71 (M<sup>+</sup> - 78), 69 (M<sup>+</sup> - 80), 67 (M<sup>+</sup> - 82), 65 (M<sup>+</sup> - 84), 63 (M<sup>+</sup> - 86), 61 (M<sup>+</sup> - 88), 59 (M<sup>+</sup> - 90), 57 (M<sup>+</sup> - 92), 55 (M<sup>+</sup> - 94), 53 (M<sup>+</sup> - 96), 51 (M<sup>+</sup> - 98), 49 (M<sup>+</sup> - 100), 47 (M<sup>+</sup> - 102), 45 (M<sup>+</sup> - 104), 43 (M<sup>+</sup> - 106), 41 (M<sup>+</sup> - 108), 39 (M<sup>+</sup> - 110), 37 (M<sup>+</sup> - 112), 35 (M<sup>+</sup> - 114), 33 (M<sup>+</sup> - 116), 31 (M<sup>+</sup> - 118), 29 (M<sup>+</sup> - 120), 27 (M<sup>+</sup> - 122), 25 (M<sup>+</sup> - 124), 23 (M<sup>+</sup> - 126), 21 (M<sup>+</sup> - 128), 19 (M<sup>+</sup> - 130), 17 (M<sup>+</sup> - 132), 15 (M<sup>+</sup> - 134), 13 (M<sup>+</sup> - 136), 11 (M<sup>+</sup> - 138), 9 (M<sup>+</sup> - 140), 7 (M<sup>+</sup> - 142), 5 (M<sup>+</sup> - 144), 3 (M<sup>+</sup> - 146), 1 (M<sup>+</sup> - 148).

Card : 11/12

for 5 min in boiling, followed by chromatography of the reaction product on 1103 (active towards I-IV) and hydrogenation over Pt (from PtO<sub>2</sub>) in ethyl alcohol gives VI, mp 120-122°C, n<sub>D</sub><sup>20</sup> 1.4760, d<sub>4</sub><sup>20</sup> 0.9228. The IR spectrum of VI shows absorption bands at 3000 (ν<sub>C-H</sub>), 1650 (ν<sub>C=O</sub>), 1600 (ν<sub>C=C</sub>), 1500 (ν<sub>C=C</sub>), 1450 (ν<sub>C=C</sub>), 1400 (ν<sub>C=C</sub>), 1350 (ν<sub>C=C</sub>), 1300 (ν<sub>C=C</sub>), 1250 (ν<sub>C=C</sub>), 1200 (ν<sub>C=C</sub>), 1150 (ν<sub>C=C</sub>), 1100 (ν<sub>C=C</sub>), 1050 (ν<sub>C=C</sub>), 1000 (ν<sub>C=C</sub>), 950 (ν<sub>C=C</sub>), 900 (ν<sub>C=C</sub>), 850 (ν<sub>C=C</sub>), 800 (ν<sub>C=C</sub>), 750 (ν<sub>C=C</sub>), 700 (ν<sub>C=C</sub>), 650 (ν<sub>C=C</sub>), 600 (ν<sub>C=C</sub>), 550 (ν<sub>C=C</sub>), 500 (ν<sub>C=C</sub>), 450 (ν<sub>C=C</sub>), 400 (ν<sub>C=C</sub>), 350 (ν<sub>C=C</sub>), 300 (ν<sub>C=C</sub>), 250 (ν<sub>C=C</sub>), 200 (ν<sub>C=C</sub>), 150 (ν<sub>C=C</sub>), 100 (ν<sub>C=C</sub>), 50 (ν<sub>C=C</sub>), 0 (ν<sub>C=C</sub>). The UV spectrum of VI shows absorption bands at 280 (ε 10000), 260 (ε 10000), 240 (ε 10000), 220 (ε 10000), 200 (ε 10000), 180 (ε 10000), 160 (ε 10000), 140 (ε 10000), 120 (ε 10000), 100 (ε 10000), 80 (ε 10000), 60 (ε 10000), 40 (ε 10000), 20 (ε 10000), 0 (ε 10000). The NMR spectrum of VI shows absorption bands at 7.5 (s, 1H), 7.0 (s, 1H), 6.5 (s, 1H), 6.0 (s, 1H), 5.5 (s, 1H), 5.0 (s, 1H), 4.5 (s, 1H), 4.0 (s, 1H), 3.5 (s, 1H), 3.0 (s, 1H), 2.5 (s, 1H), 2.0 (s, 1H), 1.5 (s, 1H), 1.0 (s, 1H), 0.5 (s, 1H), 0 (s, 1H). The mass spectrum of VI shows absorption bands at 149 (M<sup>+</sup>), 147 (M<sup>+</sup> - 2), 145 (M<sup>+</sup> - 4), 143 (M<sup>+</sup> - 6), 141 (M<sup>+</sup> - 8), 139 (M<sup>+</sup> - 10), 137 (M<sup>+</sup> - 12), 135 (M<sup>+</sup> - 14), 133 (M<sup>+</sup> - 16), 131 (M<sup>+</sup> - 18), 129 (M<sup>+</sup> - 20), 127 (M<sup>+</sup> - 22), 125 (M<sup>+</sup> - 24), 123 (M<sup>+</sup> - 26), 121 (M<sup>+</sup> - 28), 119 (M<sup>+</sup> - 30), 117 (M<sup>+</sup> - 32), 115 (M<sup>+</sup> - 34), 113 (M<sup>+</sup> - 36), 111 (M<sup>+</sup> - 38), 109 (M<sup>+</sup> - 40), 107 (M<sup>+</sup> - 42), 105 (M<sup>+</sup> - 44), 103 (M<sup>+</sup> - 46), 101 (M<sup>+</sup> - 48), 99 (M<sup>+</sup> - 50), 97 (M<sup>+</sup> - 52), 95 (M<sup>+</sup> - 54), 93 (M<sup>+</sup> - 56), 91 (M<sup>+</sup> - 58), 89 (M<sup>+</sup> - 60), 87 (M<sup>+</sup> - 62), 85 (M<sup>+</sup> - 64), 83 (M<sup>+</sup> - 66), 81 (M<sup>+</sup> - 68), 79 (M<sup>+</sup> - 70), 77 (M<sup>+</sup> - 72), 75 (M<sup>+</sup> - 74), 73 (M<sup>+</sup> - 76), 71 (M<sup>+</sup> - 78), 69 (M<sup>+</sup> - 80), 67 (M<sup>+</sup> - 82), 65 (M<sup>+</sup> - 84), 63 (M<sup>+</sup> - 86), 61 (M<sup>+</sup> - 88), 59 (M<sup>+</sup> - 90), 57 (M<sup>+</sup> - 92), 55 (M<sup>+</sup> - 94), 53 (M<sup>+</sup> - 96), 51 (M<sup>+</sup> - 98), 49 (M<sup>+</sup> - 100), 47 (M<sup>+</sup> - 102), 45 (M<sup>+</sup> - 104), 43 (M<sup>+</sup> - 106), 41 (M<sup>+</sup> - 108), 39 (M<sup>+</sup> - 110), 37 (M<sup>+</sup> - 112), 35 (M<sup>+</sup> - 114), 33 (M<sup>+</sup> - 116), 31 (M<sup>+</sup> - 118), 29 (M<sup>+</sup> - 120), 27 (M<sup>+</sup> - 122), 25 (M<sup>+</sup> - 124), 23 (M<sup>+</sup> - 126), 21 (M<sup>+</sup> - 128), 19 (M<sup>+</sup> - 130), 17 (M<sup>+</sup> - 132), 15 (M<sup>+</sup> - 134), 13 (M<sup>+</sup> - 136), 11 (M<sup>+</sup> - 138), 9 (M<sup>+</sup> - 140), 7 (M<sup>+</sup> - 142), 5 (M<sup>+</sup> - 144), 3 (M<sup>+</sup> - 146), 1 (M<sup>+</sup> - 148).

Card : 12/12 0-5

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds and Their Synthetic Analogs.

G

Abs Jour: Ref Zhur-Khim., No 11, 1959, 38777.

Author : Suchy, M. and Sorn, F.

Inst :  
Title : Chemistry of the Terpenes. XC. Synthesis of 1,7-dimethyl-4-isopropyl-cyclodecane (Germacrene)

Orig Pub: Chem Listy, 52, No 6, 1180-1185 (1958) (in Czech);  
Collection Czechoslov Chem Commun, 23, No 12, 2175-2180 (1958) (in English with a Russian summary)

Abstract: The authors have synthesized 1,7-dimethyl-4-isopropyl-cyclodecane (I) which was found to be identical with germacrene, obtained by the complete reduction of germacrene. The condensation of  $\text{CH}_3\text{CH}_2\text{COCH}_2$  with

Card : 1/7

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds and Their Synthetic Analogs

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CIA-RDP86-00513R001652420016-4"

Abs Jour: Ref Zhur-Khim., No 11, 1959, 38777.

$(\text{CH}_3)_2\text{CHCHO}$  gives the diamide of  $\alpha, \alpha'$ -dicyano- $\beta$ -isopropylglutaric acid (II), which on saponification and decarboxylation gives beta-isopropylglutaric acid (III). The diethyl ester of III on reduction with  $\text{LiAlH}_4$  gives 3-isopropylpentane-1,5-diol (IV) from which 3-isopropyl-1,5-diiodopentane (V) is obtained. The condensation of V with  $\text{CH}_2\text{CH}_2(\text{COOC}_2\text{H}_5)_2$  gives an ester which on saponification yields  $\alpha, \alpha'$ -dimethyl- $\alpha, \alpha'$ -dicarboxy- $\delta$ -isopropylazelaic acid (VI); decarboxylation of VI gives  $\alpha, \alpha'$ -dimethyl- $\delta$ -isopropylazelaic acid (VII). The hemiester of VII (VIII) [sic] is converted to the acyl chloride (E) which by the Arndt-Eistert reaction gives the dimethyl ester of  $\alpha, \beta$ -

Card : 2/7

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds and Their Synthetic Analogs. G

Abs Jour: Ref Zhur-Khin., No 11, 1959, 38777.

dimethyl- $\delta$ -isopropylsebacic acid (X). The acyloin condensation of X gives the hydroxyketone (XI). The reduction of XI according to Clemmensen gives a mixture from which I is isolated after hydrogenation of the hydrocarbon fraction. The ketone fraction is similarly converted to pure I via the ethylenethioketal and desulfurization with Raney Ni. The IR spectra of synthetic and natural I (the curves are given) are identical. 94 gms  $\text{CHCl}_2\text{CONH}_2$  in 50 ml water are treated dropwise with a mixture of 35 gms  $(\text{CH}_3)_2\text{CHCHO}$  and 4 gms piperidine and the solution is stirred for 6 hrs at  $13^\circ$ ; II is obtained, mp  $152^\circ$ . The hydrolysis of

Card : 3/7

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds and Their Synthetic Analogs. G

Abs Jour: Ref Zhur-Khin., No 11, 1959, 38777.

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II by boiling with 250 ml conc HCl gives III, mp  $102^\circ$ . The action of HCl (gas) on III in alcohol gives 36.2 gms of the diethyl ester of III, bp  $148-150^\circ/15$  mm. Reduction of the latter with  $\text{LiAlH}_4$  under the usual conditions gives IV, bp  $147-148^\circ/15$  mm. When 35 gms IV and 300 gms triphenylphosphonium methyl iodide, added in small portions, are heated 4 hrs to  $60^\circ$  and the resulting product mixture is subjected to chromatography on  $\text{Al}_2\text{O}_3$  (active towards IV), 57.1 gms of V are obtained, bp  $138-142^\circ/2$  mm. 11 gms V are added to a mixture of 11 gms  $\text{CH}_3\text{CH}(\text{CCCC}_2\text{H}_5)_2$  and 1.4 gms NaH in 70 ml toluene, the solution is heated for 1 hr at  $110^\circ$ , the tetraethyl ester is separated,

Card : 4/7

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds and Their  
Synthetic Analogs.

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Abs Jour: Ref Zhur-Khim., No 11, 1959, 38777.

saponified with methanolic KOH (10 gms per 100  
ml), the resulting solution is refluxed and treated  
[sic], giving VI, mp 176-177° (decomp; from water).  
The decarboxylation of 25.1 gms VI at 160°/15 mm  
gives 16.7 gms VII, bp 180-182°/0.1 mm. 15.7 gms  
VII in 300 ml ether are treated with 1 mol of an  
ether solution of  $\text{CH}_2\text{N}_2$  and the resulting solution  
is cooled (with stirring); 8.1 gms VIII are obtained,  
bp 145-146°/0.3 mm. 8.1 gms VIII in  $\text{C}_6\text{H}_6$  react  
with  $\text{SOCl}_2$  to give 8.7 gms IX which is added drop-  
wise with cooling to -15° to an ether solution of  
 $\text{CH}_2\text{N}_2$  (prepared from 20 gms nitrosomethylurea). The  
diazoketone which is obtained is decomposed (5 por-

Card : 5/7

CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds and Their  
Synthetic Analogs.

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Abs Jour: Ref Zhur-Khin., No 11, 1959, 38777.

pure I, bp 126°/10 mm,  $n_D^{20}$  1.4713,  $d_4^{20}$  0.8596.  
0.35 gms B of bp 149°/10 mm semicarbazone mp 165°  
(from alc) is condensed with 0.5 ml  $(CH_3SH)_2$  and  
1 ml  $BF_3 \cdot (C_2H_5)_2O$  to give the ethylenethioketal which  
on desulfurization with 5 gms of Raney Ni in 20 ml  
dioxane (24 hrs), chromatography of the hydrocarbon  
on  $Al_2O_3$ , and hydrogenation over Pt (from  $PtO_2$ ) gives  
I,  $n_D^{20}$  1.4729,  $d_4^{20}$  0.8571. -- L. Novotny.

Card : 7/7

COUNTR. : Czechoslovakia 8-3  
CATEGORY :  
ABS. JOUR. : RZhKhim., No. 16 1959, No. 52216  
AUTHOR :  
INST. :  
TITLE :  
ORIG. PUB. :  
ABSTRACT : of IV followed by oxidation with dil HNO<sub>3</sub> to D(+)-isopropylsuccinic acid (V). I does not isomerize when refluxed for 45 min with 10% NaOH in alcohol. When (-)-cadinanedihydrochloride is heated with CH<sub>3</sub>COONa in CH<sub>3</sub>COOH followed by chromatography on alkaline Al<sub>2</sub>O<sub>3</sub> followed by fractionation in a column with 70 theoretical plates packed with Diksona [sic], IV is obtained, bp 124°/9 mm, n<sub>D</sub><sup>20</sup> 1.5059, d<sub>4</sub><sup>20</sup> 0.9239. 9.5 gms IV are added over 2 hrs to

CARD: 2/6

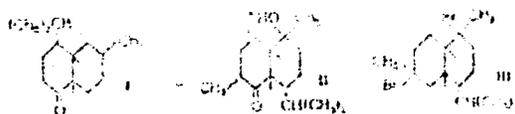
158

CARD: 3/8

ABSTRACT :  
CATION :  
ABS. JOUR. : RZKhan., No. 16 1959, No. 57:16  
AUTHOR :  
INST. :  
TITLE :

ORIG. PUB. :

ABSTRACT : benzene),  $[\alpha]_D^{20} = +20.4^\circ$  (c = 1.26; water).  
Further elution of the paper moistened with 1%  
 $H_2SO_4$  yields an additional 117 mg of V. Better



CARD: 4/6

154

COUNTRY : Czechoslovakia 7-3  
CATEGORY :  
ABS. JOUR. : AZKhim., No. 16 1959, No. 57216  
AUTHOR :  
INVT. :  
TITLE :  
ORIG. PUB. :  
ABSTRACT : yields of V are obtained by the ozonation of 7.8  
gms IV in 80 ml  $\text{CH}_3\text{COOH}$  and the decomposition of  
the ozonides at  $100^\circ$  with a mixture of 45 ml  
water and 2.2 ml 30%  $\text{H}_2\text{O}_2$ . The residue after  
the evaporation of the solution is oxidized  
(1 hr,  $110-120^\circ$ ) with 50%  $\text{HNO}_3$  and  $\text{V}_2\text{O}_5$ ; after  
the usual treatment, 2.05 gms of the anhydride  
of V are obtained which on heating with water  
give V (yield 18%); the latter is purified by  
paper chromatography. The reaction dispersion

CARD: 5/6

160

SORM F

COUNTRY : Czechoslovakia 8-3  
 CATEGORY : Organic Chemistry. Natural Compounds  
 and their Synthetic Analogues  
 ABS. JOUR. : RZKluc., No. 20 1959, No. 71529  
 AUTHOR : Schwarz, Vladimir; Cerny, Vaclav; Sorm, Frantisek  
 INST. : Not given.  
 TITLE : Steroids. XXXVII. Preparation of Methyl-3 $\beta$ ,  
 17 $\alpha$ -diacetoxybisanololcholelate and Its 20-  
 Epimer. A New Method for the Synthesis of \*  
 ORIG. PUB. : Chem. Listy, 1958, 52, #2, 1433-1441.  
 ABSTRACT : A new method for the formation of a side  
 chain in bisnor steroid acids was described.  
 The method consists of a reaction of  
 $\text{CH}_2\text{ClCH}_2$  with a corresponding 17-carbonate  
 chloroanhydride with a subsequent regrouping  
 of the resulting diazoketone. Upon the re-  
 action of Na-3 $\beta$ -acetoxy- $\Delta^5$ -etionate (I) with  
 $(\text{COCl})_2$  in  $\text{C}_6\text{H}_6$  in the presence of pyridine  
 (2 hours, 20 $^\circ$ ), 1-chloroanhydride was formed  
 (99% yield). Its solution in  $\text{C}_6\text{H}_6$  (no puri-  
 fication) was added at -20 $^\circ$  into an ether  
 solution of  $\text{CH}_2\text{ClCH}_2$ . The mixture was left  
 CARD: 1/2

\* Side Chains in Bisnor Steroid Acids

31

COUNTRY : Czechoslovakia G-3  
 CATEGORY :  
 ABS. JOUR. : RZKhim., No. 20 1959, No. 71539  
 AUTHOR :  
 INST. :  
 TITLE :  
 ORIG. PUB. :  
 ABSTRACT : #4, 11431), a mixture of three substances was obtained. Upon repeated chromatography on  $Al_2O_3$  the mixture yielded 604 mg of methyl 3 $\beta$ -acetoxy- $\Delta^5$ -biscnorcholenate (III), m.p. 136-137;  $[\alpha]_D^{25} - 62 \pm 0.5^\circ$  (from 1.8). Upon reacting 3 $\beta$ -acetoxyalloethionic acid (IV) with  $SOCl_2$  ( $0^\circ$ , followed by 16 hours at  $20^\circ$ ) IV-chloroanhydride, (97% yield) was obtained. The reaction of the latter with  $CH_3CHN_2$  (by the above-described method) yielded 17 $\beta$ -~~(diisopropionyl)-~~androstanol-3 $\beta$ -acetate (V), 85% yield, m.p. 141-143 (dec),  $[\alpha]_D^{19} - 16.6 \pm 3^\circ$  (from 1.3).

CARD: 3/8

37

COUNTRY : Czechoslovakia G-3  
 ORIGIN :  
 REF. JOUR. : RZhKhB., No. 20 1959, No. 71589  
 AUTHOR :  
 TITLE :

ORIG. NOS. :  
 ABSTRACT : a corresponding chloroanhydride (93% yield) by means of the above-described method. The latter, reacted with  $CH_3OH$ , gave an impure 17 $\alpha$ -diazopropionyl-androst-4-en-3 $\beta$ -ol-20-one-17-acetate-benzoate (VII), 91% yield. Rearrangement of VII at the above conditions, followed by saponification, esterification, acetylation and chromatographic separation over  $Al_2O_3$ , led to the formation of methyl 3 $\beta$ ,17 $\alpha$ -diacetoxybismorallolcholelate (VIII), 19% yield, m.p. 143-144 $^\circ$  (from  $CH_3OH$ ),  $[\alpha]_D^{20} = +58 \pm 1.5^\circ$  (from 1.9). After the elimination of VIII from the mother liquor

CIA ID: 576

COUNTRY : Czechoslovakia G-3  
 CATEGORY :  
 RES. JOUR. : RZKhim., No. 20 1959, No. 71589  
 AUTHOR :  
 TITLE :  
 ORIG. PUB. :  
 ABSTRACT : dithiol (2 hours, 0°C,) and introduction of dry HCl led to the formation of methyl-3-ethylsaccharobisnorallocholanate (XI), 90% yield, m.p. 168-169° (precipitation with hot CH<sub>3</sub>OH from a hot solution of CHCl<sub>3</sub>), [α]<sub>D</sub><sup>20</sup> + 20 ± 2° (from 1.1). Upon boiling of XI for 5 hours, with sponge Ni in alcohol, IX was obtained, 79% yield, m.p. 100°, [α]<sub>D</sub><sup>20</sup> + 10 ± 0.5° (from 2.4). In a similar manner 50 mg of VIII was saponified by shaking it for 16 hours with 4 ml CH<sub>3</sub>OH, containing 60 mg K<sub>2</sub>CO<sub>3</sub>. Thus obtained unpurified dioxy compound was oxidized with 30 mg CrO<sub>3</sub> in

CARD: 7/8

31

COUNTRY : Czechoslovakia G-3  
 CATEGORY :  
 RES. JOUR. : RZKhim., No. 20 1959, No. 71589  
 AUTHOR :  
 TITLE :  
 ORIG. PUB. :  
 ABSTRACT : 2 ml CH<sub>3</sub>COOH (16 hours, 20°). The resulting dithione was condensed with ethanedithiol (at the above-described conditions) and after boiling with sponge Ni and Al<sub>2</sub>O<sub>3</sub> chromatography was transformed into 12 mg of IX. All [α]<sub>D</sub><sup>20</sup> were determined in CHCl<sub>3</sub>. Article RZKhim, 1959, #15, 51547.

-- B. Anan'chenko

CARD: 8/8



COUNTRY : Czechoslovakia  
CITY/CORP :

9-9

AND. SOUR. : REKHA., No. 22 1959, No.

70719

AUTHOR :  
INST. :  
TITLE :

ORIG. PUB. :

ABSTRACT

: semicarbazone of the acetate of  $\Delta^3$ -B-norandrostenol-3  $\beta$ -one-17 (I), mp 264-266°. A solution of 0.25 gm I in 5 ml  $CH_3COOH$  and 0.7 ml of 30%  $CH_2O$  are heated for 1 hr at 65°, and the solution is poured into water and extracted with ether; 0.19 gm of the acetate of  $\Delta^3$ -B-norandrostenol-3  $\beta$ -one-17 (II ketol, mp 155-136° (from  $CH_2OH$ )) is obtained,  $[\alpha]_D^{20} -52 \pm 5^\circ$  (c = 3.1). A solution of 0.1 gm of the acetate of II in 5 ml  $CH_2OH$  is refluxed for 1 hr in a stream of  $N_2$  with

CARD: 4/8

COUNTRY : Czechoslovakia  
CALLENO : :

4-5

ABST. JOUR. : RZKhim., No. 22 1959, No.

78719

AUTHOR :  
ISSN :  
TITLE :

ORIG. PUB. :

ABSTRACT :

with ether. The acid fraction is recovered from the ether extract by extraction with 4 gms Na<sub>2</sub>CO<sub>3</sub> in 150 ml water; the addition of dil HCl to the latter extract gives 3.9 gms of 3 $\beta$ -acetoxy-5,17-diketo-5,6-Secol[Sequoi]-6-androstanic acid (IV), mp 162-163° (from ether), [ $\alpha$ ]<sub>D</sub><sup>20</sup> +70-3° (c = 3.7). The ether layer gives 2.9 gms of the acetate of  $\Delta^5$ -androstenol-3-dione-7,17, mp 185-186° (from CH<sub>2</sub>OH), [ $\alpha$ ]<sub>D</sub><sup>20</sup> -74+5° c = 2.3; in alc). A mixture of 3.5 gms IV, 9 ml pyridine,

CARD: 6/8

136



